

Name : _____ Date : _____

Acids and Bases Strength

Write True (T) or False (F).

- i. Stronger acids have weaker conjugate bases and weaker acids have stronger conjugate bases. _____
- ii. Stronger bases have stronger conjugate acids and weaker bases have weaker conjugate acids. _____

Write equations that show how the following reactions occur in water.

- i. F^- acts as a base. _____
- ii. HNO_2 acts as an acid. _____
- iii. $\text{Fe}(\text{H}_2\text{O})_6^{3+}$ acts as an acid. _____
- iv. HCO_3^- acts as a base. _____
- v. HCO_3^- acts as an acid. _____
- vi. $\text{Al}(\text{H}_2\text{O})_5(\text{OH})^{2+}$ acts as a base. _____

What volume of 0.45 M LiOH would be needed to neutralize 60.0 mL of 0.15 M HI?

Calculate the pH values of the following solutions of strong acids and bases.

- i. 0.001 M HCl
- ii. 0.76 M KOH
- iii. 2.8×10^{-4} M $\text{Ba}(\text{OH})_2$

Lecture 29 Acids And Bases I Worksheet

Clifford Lane



Lecture 29 Acids And Bases I Worksheet:

Adopting the Tune of Term: An Mental Symphony within **Lecture 29 Acids And Bases I Worksheet**

In a global consumed by screens and the ceaseless chatter of quick transmission, the melodic elegance and mental symphony developed by the written term usually disappear into the backdrop, eclipsed by the constant noise and interruptions that permeate our lives. However, set within the pages of **Lecture 29 Acids And Bases I Worksheet** an enchanting literary treasure brimming with raw thoughts, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, this charming masterpiece conducts readers on an emotional journey, well unraveling the hidden melodies and profound influence resonating within each carefully crafted phrase. Within the depths of this moving examination, we shall investigate the book is key harmonies, analyze its enthralling publishing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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