

# NOMENCLATURE PACKET

Worksheet I: Binary Ionic Compounds (representative metals) - metals from groups 1A, 2A, and 3A (1, 2, and 13) have constant charges as ions and do NOT get Roman Numerals in their names

1. Name the following ionic compounds:

- |                            |                    |
|----------------------------|--------------------|
| a. $\text{Al}_2\text{O}_3$ | Aluminum oxide     |
| b. $\text{Cs}_2\text{O}$   | Cesium oxide       |
| c. $\text{Rb}_3\text{N}$   | Rubidium nitride   |
| d. $\text{Ca}_3\text{N}_2$ | Calcium nitride    |
| e. $\text{SrSe}$           | Strontium selenide |
| f. $\text{Cs}_2\text{S}$   | Cesium sulfide     |
| g. $\text{Al}_2\text{S}_3$ | Aluminum sulfide   |
| h. $\text{LiBr}$           | Lithium bromide    |
| i. $\text{Mg}_3\text{N}_2$ | Magnesium nitride  |
| j. $\text{CaF}_2$          | Calcium fluoride   |

2. Write the chemical formula for the following ionic compounds:

- |                                |                  |                  |                         |
|--------------------------------|------------------|------------------|-------------------------|
| a. Barium nitride              | $\text{Ba}^{+2}$ | $\text{N}^{-3}$  | $\text{Ba}_3\text{N}_2$ |
| b. Indium Fluoride             | $\text{In}^{+3}$ | $\text{F}^{-1}$  | $\text{InF}_3$          |
| c. Calcium oxide               | $\text{Ca}^{+2}$ | $\text{O}^{-2}$  | $\text{CaO}$            |
| d. Sodium nitride              | $\text{Na}^{+1}$ | $\text{N}^{-3}$  | $\text{Na}_3\text{N}$   |
| e. Magnesium chloride          | $\text{Mg}^{+2}$ | $\text{Cl}^{-1}$ | $\text{MgCl}_2$         |
| f. Potassium oxide             | $\text{K}^{+1}$  | $\text{O}^{-2}$  | $\text{K}_2\text{O}$    |
| g. Magnesium oxide             | $\text{Mg}^{+2}$ | $\text{O}^{-2}$  | $\text{MgO}$            |
| h. Potassium sulfide           | $\text{K}^{+1}$  | $\text{S}^{-2}$  | $\text{K}_2\text{S}$    |
| i. Lithium nitride             | $\text{Li}^{+}$  | $\text{N}^{-3}$  | $\text{Li}_3\text{N}$   |
| j. Strontium fluoride          | $\text{Sr}^{+2}$ | $\text{F}^{-1}$  | $\text{SrF}_2$          |
| k. Aluminum sulfide            | $\text{Al}^{+3}$ | $\text{S}^{-2}$  | $\text{Al}_2\text{S}_3$ |
| l. Duplicate problem - removed |                  |                  |                         |

# Ions And Their Chemical Formula Answers

**Lei Huang**



## **Ions And Their Chemical Formula Answers:**

## Reviewing **Ions And Their Chemical Formula Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Ions And Their Chemical Formula Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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