POGIL 2 Ions

How are lone made from neutral atoms?

You have teamed that not all atoms of an element are the same. Variation in the number of neutrons results in different isotopes of the element. In this activity we will explore another variation that can take place—the loss and gain of electrons. The exchange of electrons between atoms is a very common way for chemical change to take place. We will see it many times throughout the year.

Use Disserain 1 or the rest asset to complete the following: table.

	Metal or Nonmetal	Number of pretens inthe store	Wassber of protons in the ion	Mumber of neutrons in the attent	Number of secotroes in the ion	Number of electrons in the atom	Number of electrons in the item	Change on the lon
Lithium:	rected	3		4	-	3	2	13%
Magnissians	Metal	12	7.2	7.2	12	12	1.0	200
Akaminum	Metalloid	13	7.3	14	34	15	10	311
Physicae	Non-metal	9	and the same	10	10	0	1.0	100
Ovegen	(homestall)	8	N		.81	8	1.0	2-
Nitrogen	Non-meter	7	7	7	7	7	5.0	3-

Based on the table you completed in Question 1, what distinguishes a neutral atom flore an ion?
 A weatral atom has the same number of protons and electrons (no charge). An ion has a charge (it of protons does not equal it of electrons).

- 3. Examine the isstope symbols in Diagram I...
 - a. Where is the lan charge located in the leasupe symbol?

Top right corner

- b. Is a charge indicated on the neutral atoms? If yes, where is it located? No.
- Which substrettés particle carries a positive charge?

Protons are positive

5. Which subspenie particle carries a negative charge?

Electrons are negative

 Propose a mathematical equation to calculate the change on an ion from the number of presons and electrons in an ion. Confirm that your equation works using two posts valors examples and two possitive ion examples from Dissersen 1.

The charge of an ion - # of protons - # of electrons

Ions Pogil Packet Answer Key

Minjie Lin

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Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more

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web align the timing marks of the camshaft with the mark plates of the timing chain and install the timing chain hint align the timing marks with the mark plates while turning the hexagonal service portion of the camshaft using a wrench 2 install chain tensioner slipper a install the chain tensioner slipper 3 install chain tensioner

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web rotate the crankshaft pulley two complete turns clockwise and check the timing marks if the timing marks are correct tighten the lock bolt for the tensioner to 20 to 34 n m 200 to 350 kgfcm for fe to 37 to 52 n m 380 to 530 kgfcm for f2 engine check the movement on the tension side of the timing belt

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web jul 22 2009 a set the timing mark to 0 on the timing chain cover by rotating the crankshaft clockwise b verify the timing marks on the camshaft timing gears are in general alignment with the timing marks of the camshaft bearing cap see chart note when the engine is correctly timed the timing marks will not perfectly align

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