

Ionic Chemical Formulas/PreAICE Chemistry

NAME: _____ PERIOD _____

Directions: Using the Chemical Formulas program sheet, fill in the table below. Use subscripts & superscripts. Remember to use roman numerals to show the valence of the transition metals in the compound name.

| METAL CATION (+) | CATION FORMULA | NONMETAL ANION (-) | ANION FORMULA | COMPOUND FORMULA | COMPOUND NAME |
|------------------|------------------------------|--------------------|-------------------------------|---|-----------------------|
| sodium | Na ⁺¹ | hydroxide | OH ⁻ | NaOH | sodium hydroxide |
| magnesium | Mg ⁺² | phosphate | PO ₄ ⁻³ | Mg ₃ (PO ₄) ₂ | magnesium phosphate |
| calcium | Ca ⁺² | chloride | Cl ⁻ | CaCl ₂ | calcium chloride |
| gold(I) | Au ⁺¹ | sulfite | SO ₃ ⁻² | Au ₂ SO ₃ | gold(I) sulfite |
| copper(II) | Cu ⁺² | hydroxide | OH ⁻ | Cu(OH) ₂ | copper(II) hydroxide |
| Iron(III) | Fe ⁺³ | telluride | Te ⁻² | Fe ₂ Te ₃ | iron(III) telluride |
| magnesium | Mg ⁺² | bromide | Br ⁻¹ | MgBr ₂ | magnesium bromide |
| manganese(II) | Mn ⁺² | sulfide | S ⁻² | MnS | manganese(II) sulfide |
| boron | B ⁺³ | oxide | O ⁻² | B ₂ O ₃ | boron oxide |
| potassium | K ⁺ | nitrate | NO ₃ ⁻ | KNO ₃ | potassium nitrate |
| barium | Ba ⁺² | bromide | Br ⁻ | BaBr ₂ | barium bromide |
| strontium | Sr ⁺² | sulfate | SO ₄ ⁻² | SrSO ₄ | strontium sulfate |
| iron(III) | Fe ⁺³ | oxide | O ⁻² | Fe ₂ O ₃ | iron(III) oxide |
| silver(I) | Ag ⁺¹ | chloride | Cl ⁻ | AgCl | silver(I) chloride |
| aluminum | Al ⁺³ | carbonate | CO ₃ ⁻² | Al ₂ (CO ₃) ₃ | aluminum carbonate |
| gold(II) | Au ⁺² | nitride | N ⁻³ | Au ₃ N ₂ | gold(II) nitride |
| cesium | Cs ⁺¹ | phosphate | PO ₄ ⁻³ | Cs ₃ PO ₄ | cesium phosphate |
| ammonium | NH ₄ ⁺ | sulfide | S ⁻² | (NH ₄) ₂ S | ammonium sulfide |
| zinc(II) | Zn ⁺² | nitrite | NO ₂ ⁻ | Zn(NO ₂) ₂ | zinc(II) nitrite |
| chromium(I) | Cr ⁺¹ | nitride | N ⁻³ | Cr ₃ N | chromium(I) nitride |
| mercury(II) | Hg ⁺² | oxide | O ⁻² | HgO | mercury(II) oxide |
| sodium | Na ⁺ | sulfite | SO ₃ ⁻² | Na ₂ SO ₃ | sodium sulfite |
| magnesium | Mg ⁺² | oxide | O ⁻² | MgO | magnesium oxide |

Ions Answer Key

G Orfield

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'G Orfield' text.

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