

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Number of Electrons Lost or Gained	Charge
1	Sodium	Na^+	11	10	lost one	positive
2	Chlorine	Cl^-	17	18	gained one	negative
3	Calcium	Ca^{2+}	20	18	lost two	positive
4	Oxygen	O^{2-}	8	10	gained two	negative
5	Aluminum	Al^{3+}	13	10	lost three	positive
6	Sulfur	S^{2-}	16	18	gained two	negative
7	Hydrogen	H^+	1	0	lost one	positive
8	Lithium	Li^+	3	2	lost one	positive
9	Fluorine	F^-	9	10	gained one	negative
10	Magnesium	Mg^{2+}	12	10	lost two	positive
11	Strontium	Sr^{2+}	38	36	lost two	positive
12	Barium	Ba^{2+}	56	54	lost two	positive
13	Iron	Fe^{2+}	26	24	lost two	positive
14	Copper	Cu^+	29	28	lost one	positive
15	Gold	Au^+	79	78	lost one	positive
16	Mercury	Hg^{2+}	80	78	lost two	positive
17	Lead	Pb^{2+}	82	80	lost two	positive
18	Platinum	Pt^{2+}	78	76	lost two	positive
19	Silver	Ag^+	47	46	lost one	positive
20	Vanadium	V^{3+}	23	20	lost three	positive
21	Chromium	Cr^{3+}	24	21	lost three	positive
22	Manganese	Mn^{2+}	25	23	lost two	positive
23	Zinc	Zn^{2+}	30	28	lost two	positive
24	Cadmium	Cd^{2+}	48	46	lost two	positive
25	Barium	Ba^{2+}	56	54	lost two	positive
26	Strontium	Sr^{2+}	38	36	lost two	positive
27	Calcium	Ca^{2+}	20	18	lost two	positive
28	Magnesium	Mg^{2+}	12	10	lost two	positive
29	Sodium	Na^+	11	10	lost one	positive
30	Lithium	Li^+	3	2	lost one	positive
31	Hydrogen	H^+	1	0	lost one	positive
32	Helium	He^0	2	2	none	neutral
33	Lithium	Li^+	3	2	lost one	positive
34	Beryllium	Be^{2+}	4	2	lost two	positive
35	Boron	B^{3+}	5	2	lost three	positive
36	Carbon	C^{4+}	6	2	lost four	positive
37	Nitrogen	N^{3-}	7	10	gained three	negative
38	Oxygen	O^{2-}	8	10	gained two	negative
39	Fluorine	F^-	9	10	gained one	negative
40	Neon	Ne^0	10	10	none	neutral
41	Sodium	Na^+	11	10	lost one	positive
42	Magnesium	Mg^{2+}	12	10	lost two	positive
43	Aluminum	Al^{3+}	13	10	lost three	positive
44	Silicon	Si^{4-}	14	18	gained four	negative
45	Phosphorus	P^{3-}	15	18	gained three	negative
46	Sulfur	S^{2-}	16	18	gained two	negative
47	Chlorine	Cl^-	17	18	gained one	negative
48	Argon	Ar^0	18	18	none	neutral
49	Potassium	K^+	19	18	lost one	positive
50	Calcium	Ca^{2+}	20	18	lost two	positive
51	Gallium	Ga^{3+}	31	28	lost three	positive
52	Germanium	Ge^{4+}	32	28	lost four	positive
53	Antimony	Sb^{3-}	33	36	gained three	negative
54	Tellurium	Te^{2-}	34	36	gained two	negative
55	Iodine	I^-	53	54	gained one	negative
56	Xenon	Xe^0	54	54	none	neutral
57	Cesium	Cs^+	55	54	lost one	positive
58	Barium	Ba^{2+}	56	54	lost two	positive
59	Lanthanum	La^{3+}	57	54	lost three	positive
60	Cerium	Ce^{4+}	58	54	lost four	positive
61	Praseodymium	Pr^{3+}	59	56	lost three	positive
62	Nd	Nd^{3+}	60	57	lost three	positive
63	Promethium	Pm^{3+}	61	58	lost three	positive
64	Samarium	Sm^{2+}	62	60	lost two	positive
65	Europium	Eu^{2+}	63	61	lost two	positive
66	Gadolinium	Gd^{3+}	64	61	lost three	positive
67	Terbium	Tb^{3+}	65	62	lost three	positive
68	Dysprosium	Dy^{3+}	66	63	lost three	positive
69	Ho	Ho^{3+}	67	64	lost three	positive
70	Er	Er^{3+}	68	65	lost three	positive
71	Tm	Tm^{3+}	69	66	lost three	positive
72	Yb	Yb^{2+}	70	68	lost two	positive
73	Lu	Lu^{3+}	71	68	lost three	positive
74	Hafnium	Hf^{4+}	72	68	lost four	positive
75	Tantalum	Ta^{5+}	73	68	lost five	positive
76	Tungsten	W^{6+}	74	68	lost six	positive
77	Rhenium	Re^{7+}	75	68	lost seven	positive
78	Osmium	Os^{8+}	76	68	lost eight	positive
79	Iridium	Ir^{9+}	77	68	lost nine	positive
80	Platinum	Pt^{2+}	78	76	lost two	positive
81	Gold	Au^{3+}	79	76	lost three	positive
82	Mercury	Hg^{2+}	80	78	lost two	positive
83	Thallium	Tl^{3+}	81	78	lost three	positive
84	Lead	Pb^{2+}	82	80	lost two	positive
85	Bismuth	Bi^{3+}	83	80	lost three	positive
86	Polonium	Po^{2+}	84	82	lost two	positive
87	Astatine	At^{3-}	85	88	gained three	negative
88	Radon	Rn^0	86	86	none	neutral
89	Francium	Fr^+	87	86	lost one	positive
90	Radium	Ra^{2+}	88	86	lost two	positive
91	Actinium	Ac^{3+}	89	86	lost three	positive
92	Thorium	Th^{4+}	90	86	lost four	positive
93	Protactinium	Pa^{5+}	91	86	lost five	positive
94	Uranium	U^{6+}	92	86	lost six	positive
95	Np	Np^{3+}	93	90	lost three	positive
96	Pu	Pu^{4+}	94	90	lost four	positive
97	Americium	Am^{3+}	95	92	lost three	positive
98	Cm	Cm^{3+}	96	93	lost three	positive
99	Berkelium	Bk^{3+}	97	94	lost three	positive
100	Californium	Cf^{3+}	98	95	lost three	positive
101	Einsteinium	Es^{3+}	99	96	lost three	positive
102	Fermium	Fm^{2+}	100	98	lost two	positive
103	Mendelevium	Md^{3+}	101	98	lost three	positive
104	Nobelium	No^{2+}	102	100	lost two	positive
105	Lawrencium	Lr^{3+}	103	100	lost three	positive
106	Rutherfordium	Rf^{4+}	104	100	lost four	positive
107	Dubnium	Db^{5+}	105	100	lost five	positive
108	Seaborgium	Sg^{6+}	106	100	lost six	positive
109	Bohrium	Bh^{7+}	107	100	lost seven	positive
110	Hassium	Hs^{8+}	108	100	lost eight	positive
111	Meitnerium	Mt^{9+}	109	100	lost nine	positive
112	Darmstadtium	Ds^{10+}	110	100	lost ten	positive
113	Roentgenium	Rg^{9+}	111	100	lost nine	positive
114	Copernicium	Cn^{8+}	112	100	lost eight	positive
115	Nihonium	Nh^{7+}	113	100	lost seven	positive
116	Flerovium	Fl^{6+}	114	100	lost six	positive
117	Moscovium	Mc^{5+}	115	100	lost five	positive
118	Livermorium	Lv^{4+}	116	100	lost four	positive
119	Tennessine	Ts^{3+}	117	100	lost three	positive
120	Oganesson	Og^{2+}	118	100	lost two	positive

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