



# CHEM FAX!

FLINN  
LABORATORY EQUIPMENT

## Preparing and Diluting Solutions Student Laboratory Kit

### Introduction

Solutions are homogeneous mixtures of two or more substances. The components of a solution are called solutes and solvents. A solution is a homogeneous mixture of two or more substances. The components of a solution are called solutes and solvents.

### Concepts

- 1. Solubility
- 2. Concentration

### Background

#### Preparing Solutions

The amount of solute that can be dissolved in a given amount of solvent is called the solubility of the solute in the solvent. The solubility of a substance in a solvent is a characteristic property of the substance and the solvent.

When a substance is dissolved in a solvent, the resulting mixture is called a solution. The concentration of a solution is the amount of solute present in a given amount of solution.

#### Concentration

Concentration is the amount of solute present in a given amount of solution. It is usually expressed in terms of mass per volume (m/v) or mass per mass (m/m).

Mass per volume (m/v) is the most common way to express concentration.

Mass per mass (m/m) is used when the mass of the solution is known.

Volume per volume (v/v) is used when the volume of the solution is known.

Temperature per volume (t/v) is used when the temperature of the solution is known.

Pressure per volume (p/v) is used when the pressure of the solution is known.

Concentration is a property of a solution and is independent of the amount of solution.

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# Flinn Scientific Preparing And Diluting Answers

**David Baud**



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