

**CONDENSED CONSOLIDATED STATEMENT OF FINANCIAL POSITION**

**AT 31 DECEMBER, 2010**

<b>ASSETS</b>	<b>2010</b>	<b>2009</b>	<b>2008</b>	<b>Equity and Liabilities</b>	<b>2010</b>	<b>2009</b>	<b>2008</b>
	<b>\$ million</b>	<b>\$ million</b>	<b>\$ million</b>		<b>\$ million</b>	<b>\$ million</b>	<b>\$ million</b>
<b>ASSETS</b>				<b>EQUITY</b>			
<b>Non-current assets</b>				Equity attributable to owners of the Company	62,489	59,352	56,546
Property, plant and equipment	28,781	25,053	19,405	Non-controlling interests	6,000	4,988	4,173
Deposits for acquisition of property, plant and equi	1,198	1,597	916	<b>Total equity</b>	<b>68,489</b>	<b>64,340</b>	<b>60,719</b>
Lease prepayments	6,946	6,216	6,106				
Deposits for land use rights	121	87	14	<b>LIABILITIES</b>			
Deposits for investment	141	-	-	<b>Non-current liabilities</b>			
Investment properties	1,705	1,548	890	Other payables	204	1,158	475
Intangible assets	19,923	15,309	8,804	Borrowings	35,347	20,676	13,648
Mining assets	2,254	1,342	47	Obligations under finance lease	118	394	391
Interests in jointly controlled entities	935	742	639	Financial guarantee contracts	32	35	77
Interests in associates	3,680	3,679	3,362	Retirement and other supplemental benefit obligati	6,519	6,421	8,323
Goodwill	836	836	897	Provisions	46	48	0
Available-for-sale financial assets	4,661	4,887	1,807	Deferred income government grant	406	157	215
Other loans and receivables	3,502	884	1,336	Deferred income tax liabilities	482	453	162
Deferred income tax assets	2,794	2,273	1,905		<b>43,154</b>	<b>29,342</b>	<b>23,291</b>
Other prepayments	57	40	37				
Trade and other receivables	1,895	614	62	<b>Current liabilities</b>			
	<b>79,429</b>	<b>65,127</b>	<b>46,327</b>	Trade and other payables	181,255	136,756	112,933
<b>Current assets</b>				Amounts due to customers for contract work	13,985	10,194	2,484
Lease prepayments	162	108	121	Current income tax liabilities	612	583	589
Properties held for sale	4,133	1,877	620	Borrowings	26,110	39,409	33,665
Properties under development for sale	27,103	20,343	14,779	Obligations under finance lease	296	258	60
Inventories	26,158	21,349	14,643	Financial guarantee contracts	2	1	2
Trade and other receivables	109,178	80,782	67,420	Retirement and other supplemental benefit obligati	777	856	325
Amounts due from customers for contract work	39,076	36,630	36,169	Held-for-trading financial liabilities	121	86	0
Other loans and receivables	1,709	949	394		<b>229,158</b>	<b>188,143</b>	<b>150,038</b>
Held-for-trading financial assets	124	101	160				
Restricted cash	2,239	2,808	2,115	<b>Total liabilities</b>	<b>266,312</b>	<b>217,485</b>	<b>173,329</b>
Cash and cash equivalents	45,490	51,651	51,400				
	<b>255,372</b>	<b>216,698</b>	<b>187,821</b>	<b>Total equity and liabilities</b>	<b>334,801</b>	<b>281,825</b>	<b>234,048</b>
<b>Total assets</b>	<b>334,801</b>	<b>281,825</b>	<b>234,048</b>				
				Net current assets	32,214	20,555	37,783
				Total assets less current liabilities	111,643	93,682	84,050

# Financial Report Of First

**L Cohen**



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web donor whereas a base generically represented by  $b$  is a proton acceptor when acids and bases combine they swap a proton  $h_a$   $b + h_b$  strong acids e.g. sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the  $K_a$  or  $K_b$  from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral pH at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web 1 approximately 3 g of hydrated oxalic acid  $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$  was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of hcl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the ph of the solution is then  $\text{ph} = -\log 0.100 = 1.000$  ph log 0.100 1.000 b titrant volume 12.50 ml

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web one type of titration uses a neutralization reaction in which an acid and a base react to produce a salt and water in equation 1 the acid is hcl hydrochloric acid and the base is naoh sodium hydroxide when the acid and base react they form nacl sodium chloride which is also known as table salt

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web for example the scaa brewing standard for preparation of drip coffee defines the proper ratios of water to coffee the proper extraction brewing temperature and holding

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web the ratio of coffee to water when preparing a brew in this standard brew ratio is expressed in grams of coffee per 1 000 kg of fresh water at 17 3 c g kg the brew

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