

Worksheet – Calorimetry

Calorimetry is the experimental measurement of heat (q) produced in chemical and physical processes. Heat can not be measured directly, but temperature changes can be measured. The factor that links these two is **heat capacity**.

Heat capacity, C , is defined as the heat required to raise the temperature of a substance one degree. It relates a measured change in temperature (ΔT) to heat flow (q) by the equation:

$$q = C \times \Delta T \times \text{mass}$$

This can be calculated on a per gram basis (specific heat capacity) or a per mole basis (molar heat capacity).

It is easiest to understand the molecular basis for heat capacity by using molar heat capacities of different substances. As energy enters a system, it is distributed among all of the degrees of freedom within a molecule (translational, rotational and vibrational modes). The more complex the molecule, the larger its heat capacity, and the smaller the rise in temperature, since the energy is spread out among more modes.

1. 10 gram samples of iron at a temperature of 100°C are placed into three beakers each containing 1.0 mol of three different solvents, octane, trichloromethane and methanol, all at 25.0°C . Their structures and molar masses are shown below.



octane C_8H_{18}
($\text{mw} = 114 \text{ g/mol}$)



trichloromethane CHCl_3
($\text{mw} = 119 \text{ g/mol}$)



methanol CH_3OH
($\text{mw} = 32 \text{ g/mol}$)

- How many degrees of freedom in each compound? $3 \times n$ $n = \# \text{ atoms}$
 octane $26 \times 3 = 78$ trichloromethane $3 \times 5 = 15$ methanol $3 \times 6 = 18$
- Which will have the largest molar heat capacity? *most modes - octane*
- This compound have the (largest, smallest) change in temperature.
- Which will have the largest change in temperature? *smallest C = trichloro methane*
- The temperature of the methanol rises to 28.9°C . The specific heat capacity of iron is $0.449 \text{ J/g}^\circ\text{C}$. What are the **specific and molar heat capacities** of methanol?
 $q_{\text{metal}} = -q_{\text{solvent}}$ $1 \text{ mol} = 32 \text{ g}$

$$\begin{aligned}
 (0.449 \text{ J/g}^\circ\text{C})(10 \text{ g})(28.9 - 100) &= -(C)(32 \text{ g})(28.9 - 25) \\
 C &= \frac{2.56 \text{ J}}{\text{g}^\circ\text{C}} \times \frac{32 \text{ g}}{\text{mol}} = 81.9 \text{ J/mol}^\circ\text{C}
 \end{aligned}$$

Energy Pogil Answers Calorimetry

Richard B. Kemp

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Calorimetry Juan Carlos Moreno Piraján, 2018-07-18 Today calorimetry is considered an art although some consider it a tool that studies the energy changes that occur during a change of state This allows physicochemical analysis to study in detail the thermodynamic systems and to evaluate the different variables that establish the characteristics of the system itself This book illustrates how the reader can use this technique in a wide spectrum of applications *Calorimetry for Collider Physics, an Introduction* Michele Livan, Richard Wigmans, 2019-07-09 This book is exceptional in offering a thorough but accessible introduction to calorimetry that will meet the needs of both students and researchers in the field of particle physics It is designed to provide the sound knowledge of the basics of calorimetry and of calorimetric techniques and instrumentation that is mandatory for any physicist involved in the design and construction of large experiments or in data analysis An important feature is the correction of a number of persistent common misconceptions Among the topics covered are the physics and development of electromagnetic showers electromagnetic calorimetry the physics and development of hadron showers hadron calorimetry and calibration of a calorimeter Two chapters are devoted to more promising calorimetric techniques for the next collider *Calorimetry for Collider Physics* an introduction will be of value for all who are seeking a reliable guide to calorimetry that occupies the middle ground between the brief chapter in a generic book on particle detection and the highly complex and lengthy reference book Calorimetry Stefan Mathias Sarge, Günther W. H. Höhne, Wolfgang Hemminger, 2014-02-25 Clearly divided into three parts this practical book begins by dealing with all fundamental aspects of calorimetry The second part looks at the equipment used and new developments The third and final section provides measurement guidelines in order to obtain the best results The result is optimized knowledge for users of this technique supplemented with practical tips and tricks *Handbook of Thermal Analysis and Calorimetry* Richard B. Kemp, 1999-12-13 The applications and interest in thermal analysis and calorimetry have grown enormously during the last half of the 20th century These techniques have become indispensable in the study of processes such as catalysis hazards evaluation etc and in measuring important physical properties quickly conveniently and with markedly improved accuracy Consequently thermal analysis and calorimetry have grown in stature and more scientists and engineers have become at least part time practitioners People new to the field therefore need a source of information describing the basic principles and current state of the art The last volume of this 4 volume handbook devoted to many aspects of biological thermal analysis and calorimetry completes a comprehensive review of this important area All chapters have been prepared by recognized experts in their respective fields The approach taken is how and what to do and when to do it The complete work is a valuable addition to the already existing literature **Calorimetry in Medicine** William Sharp McCann, 1924

Calorimetry R. Wigmans, 2017 Particle physics experiments rely increasingly on calorimetry Calorimeters are intricate instruments their performance characteristics depend on design details This is a fundamental and systematic introduction

Calorimetry Richard Wigmans, 2017-10-06 Particle physics is the science that pursues the age old quest for the innermost structure of matter and the fundamental interactions between its constituents Modern experiments in this field rely increasingly on calorimetry a detection technique in which the particles of interest are absorbed in the detector Calorimeters are very intricate instruments Their performance characteristics depend on subtle sometimes counter intuitive design details Written by one of the world s foremost experts *Calorimetry* is the first comprehensive text on this topic It provides a fundamental and systematic introduction to calorimetry It describes the state of the art in terms of both the fundamental understanding of calorimetric particle detection and the actual detectors that have been or are being built and operated in experiments The last chapter discusses landmark scientific discoveries in which calorimetry has played an important role This book summarizes and puts into perspective the work described in some 900 scientific papers listed in the bibliography This second edition emphasizes new developments that have taken place since the first edition appeared in 2000

Proceedings of the Tenth International Conference on Calorimetry in Particle Physics Ren-yuan Zhu, 2003-01-16 Annotation The International Conference on Calorimetry in Particle Physics has become the major forum for state of the art developments of calorimetry techniques The tenth conference was attended by about 150 physicists from 20 countries and covered all aspects of calorimetric particle detection and measurements with emphasis on high energy physics experiments as well as experiments in nuclear physics and astrophysics The proceedings contain three parts introductory papers contributed papers and a summary The introductory papers start with a historical review of the development of calorimetry technology and continue with overviews of the current status of calorimetry in high energy physics and astrophysics which are followed by discussions on calorimetry in future accelerator facilities such as linear colliders and the Super B Factory A hot technology regarding the energy flow concept is also dealt with

Calorimetry In High Energy Physics - Proceedings Of The Fifth International Conference Howard Gordon, Doris Rueger, 1995-09-26 The Fifth International Conference on Calorimetry in High Energy Physics was held Sept 25 Oct 1 1994 at Brookhaven National Laboratory The results presented show that calorimetry is a key element in the experiments at the frontier As these experiments evolve there are new challenges for calorimetry in terms of performance in energy and position resolution at ever increasing rates The proceedings document the state of the art in calorimetry

Investigations of Structure - Energy Relationships by Reaction Calorimetry Eric John Martin, 1984

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