

EMPIRICAL AND MOLECULAR FORMULA

- 1) What is the empirical formula of a compound that contains 46.3 % lithium and 53.7% oxygen?
- 2) What is the empirical formula of a compound that contains 15.9 % boron and 84.1 % fluorine?
- 3) Phosphorus reacts with oxygen to give a compound that is 43.7% phosphorus and 56.4% oxygen. What is the empirical formula of the compound?
- 4) An inorganic salt is composed of 17.6% sodium, 39.7% chromium and 42.8% oxygen. What is the empirical formula of this salt?
- 5) Compound X contains 69.9% carbon, 6.86% hydrogen and 23.3% oxygen. Determine the empirical formula of X.
- 6) Oxalic acid has the empirical formula CHO_2 . Its molar mass is 90 g/mol. What is the molecular formula of oxalic acid?
- 7) The empirical formula of codeine is $\text{C}_{18}\text{H}_{21}\text{NO}_3$. If the molar mass of codeine is 299 g/mol, what is its molecular formula?
- 8) A compound's molar mass is 240.28 g/mol. Its percentage composition is 75.0% carbon, 5.05% hydrogen and 20.0% oxygen. What is the compound's molecular formula?
- 9) The wintergreen plant produces methyl salicylate, or oil of wintergreen. It can also be prepared easily in the laboratory. Methyl salicylate is 63.1% carbon, 5.31% hydrogen and 31.6% oxygen. Calculate the empirical formula of this compound.
- 10) An inorganic salt is made up of 38.8% calcium, 20.0% phosphorus, and 41.2% oxygen.
 - a) What is the empirical formula of the compound?
 - b) On further analysis, each formula unit of this salt is found to contain two phosphate ions. Predict the molecular formula of this salt.

Answers

- 1) Li_2O
- 2) BF_3
- 3) P_2O_5
- 4) $\text{Na}_2\text{Cr}_2\text{O}_7$
- 5) $\text{C}_{12}\text{H}_{14}\text{O}_3$
- 6) $\text{C}_2\text{H}_2\text{O}_4$
- 7) $\text{C}_{18}\text{H}_{21}\text{NO}_3$
- 8) $\text{C}_{15}\text{H}_{12}\text{O}_3$
- 9) $\text{C}_8\text{H}_8\text{O}_3$
- 10) a) $\text{Ca}_3\text{P}_2\text{O}_8$ b) $\text{Ca}_3(\text{PO}_4)_2$

Empirical And Molecular Formula Problems

Lauren Gardner



Empirical And Molecular Formula Problems:

Unveiling the Magic of Words: A Report on "**Empirical And Molecular Formula Problems**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Empirical And Molecular Formula Problems**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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