

MTI LADY'S STANDARD

Cosmetology Study Guide

The Essential Companion



Milady Essential Companion Study Guide Answer Key

Scott C. Dulebohn



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web a write a balanced equation for the reaction $\text{Al} + \text{O}_2 \rightarrow \text{Al}_2\text{O}_3$ 4 Al 3 O_2 2 Al_2O_3 b determine the theoretical yield of Al_2O_3 160.0 g Al $\times \frac{1 \text{ mol Al}}{27 \text{ g Al}} \times \frac{2 \text{ mol Al}_2\text{O}_3}{4 \text{ mol Al}} \times 101.964 \text{ g Al}_2\text{O}_3 = 302.3 \text{ g Al}_2\text{O}_3$ 26.982 g Al 4 mol Al 1 mol Al_2O_3 c determine the percent yield yield actual $\times 100$ 260.0 g $\times 100$ 86.01 theoretical 302.3 g

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