HSC Maharashtra Board question paper: March 2013

Note:

- All question are compulsory
- ii. Answer to the two sections are to be written in the same answer book.
- iii. Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Encery new question must be started on a new page.
- vi. Use of logarithmic table is allowed

CHEMISTRY: SECTION - I

Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7

i. Jo	In body centred cubic structure the space occupied is about				
0	A)	68 %	CI	3)	53 %
.0	C)	38 %	a	D)	32 %
ii. F	For a gaseous reaction the unit of rate of reaction is				
(.	A)	L atan s-1	a	3)	atm mol ⁻¹ s ⁻¹
(4	C)	atm s 1	a	((C	mol s
iii. V	Which of the following compounds contains S = O as well as S = S bonds?				
6	A)	Sulphuric acid	G	3)	Thiosulphuric acid
(4	C)	Sulphurous acid	O a	D)	Thiosulphurous acid
iv: V	Which of the following solutions shows maximum depression in freezing point?				
€.	A)	0.5 M Li ₂ SO ₁	cr.	B)	1 M NaCl
0	C)	0.5 M Al ₂ (SO ₄) ₅	σ	D)	0.5 M BaCl ₂
v. F	For a chemical reaction $\Delta S = -0.035$ kJ/K and $\Delta H = -20$ kJ.				
0	A)	5.14 K	O	B)	57.14 K
- 0	C)	571.4 K	a	D)	5714.0 K
vi. I	The standard e.m.f of the following cell is 0.463 V				
0	Cu Cu ^{**} ₍₃₅₀₎ Ag [*] ₍₃₅₀₎ Ag. If E ⁰ _{As} = 0.800 V,				

vii. Fe₂O₃ is reduced to spongy iron near the top of blast furance by

(A) H₂

(A) 1.137 V

(C) 0.463 V

(B) CaO

0.337 V

-0.463 V

(B)

(ID)

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