

PRACTICE TEST-3

TOPIC -3 (Bonding and structure)

Name: _____
Grade: XI
UNIT: 1
Time: 1Hr

DATE: .08.2024
Subject: Chemistry
Marks: 25
Invigilator sign: _____

Instructions

- Use black ink or ball-point pen.
- Fill in the boxes at the top of this page with your name, Centre number and candidate number.
- Answer all questions.
- Answer the questions in the spaces provided
- there may be more space than you need.

SECTION –A1.

1. Which of the following statements about the properties associated with ionic and covalent compounds is correct?
- A. An ionic compound cannot undergo electrolysis.
 - B. The only covalently bonded substances with high melting temperatures are those in which hydrogen bonds are present.
 - C. A compound cannot contain both ionic bonding and covalent bonding.
 - D. Ionic compounds differ from metals because they do not conduct electricity in the solid state.

Particle	Relative charge	Relative mass
proton	+1	1
neutron		
electron		

(Total for Question 1 = 2 mark)

2. This is a question about atoms, isotopes and ions.
- (a) Which of the following pairs of ions is isoelectronic?

(1)

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