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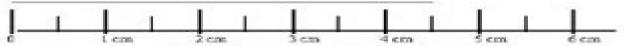
Part One: Sig Figs, Scientific Notation, SI Units, & Measurement

- 1. Significant Figures: know the rules of significant figures for:
 - a. Addition/subtraction: the answer can be no more significant than the weakest number
 - Multiplication/Division: the answer should have the same number of significant digits as the number with the smallest number of sig figs.
- Precision vs. Accuracy: Repeated measurements that are in agreement are precise. A measurement that matches the actual number is accurate.



 Uncertainty in Measurement: When reading an instrument, you are only able to estimate ONE DIGIT beyond the smallest increment.

For example: On the ruler below, the measurement would be 4.50 cm, where the 0 is estimated as one digit beyond the smallest increment.



- Percent Error: Percent Error = ((Accepted Experimental)/Accepted) x 100%
 Note: Percent Error does not have a sign, since it is based on the absolute value of the difference.
- 5. SI Unit Conversions with Factor Labeling:
 - a. Know how to convert the SI Units for length, mass, & volume based on the following prefixes:

Prefix:	Symbol	Amount
Mega	м	1,000,000 (104)
Kilo	K	1,000 (107)
Hector	н	100 (104)
Deca	Da	10 (10)
Deci	D	0.1 (101)
Centi	С	0.01 (102)
Mills	M	0.001 (10-3)
Micro	U	0.000001 (104)
Nano	N	0.000000001 (10%)

- Convert Celsius, Fahrenheit, and Kelvin temperatures.
- c. Compare/calculate mass vs. weight (with gravity)

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