

- ## SECTION C

Answer any two questions.

Question 8

- $$\text{HC}=\text{CH} \xrightarrow[\text{Hg}^{2+}]{\text{dil. H}_2\text{SO}_4} [\text{A}] \xrightarrow[\text{NH}_2\text{OH}]{[\text{O}]} [\text{B}] \xrightarrow{\text{SOCl}_2} [\text{C}] \xrightarrow[\text{C}_6\text{H}_5\text{OH}]{\text{CH}_3\text{COONa}, \Delta} [\text{D}]$$
- $$[\text{C}] \xrightarrow{\Delta} [\text{F}]$$

Question 9

- (a) Write balanced chemical equations for the following reactions and name the reactions: [3]
(i) Acetamide is heated with bromine and sodium hydroxide solution.
(ii) Benzaldehyde is treated with 50% sodium hydroxide solution.
- (b) Give one chemical test to distinguish between the following pairs of compounds: [3]
(i) Acetone and phenol.
(ii) Formic acid and Acetic acid.

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