

Chemistry of Life

Study Guide A

Answer Key

SECTION 1. ATOMS, IONS, AND MOLECULES

1. nucleus: dense center of an atom
2. neutron: particle with no electrical charge
3. proton: particle with positive electrical charge
4. electron: particle with negative electrical charge
5. compounds
6. elements
7. false
8. false
9. true
10. true
11. outermost
12. strong
13. electrons
14. covalent
15. compound
16. element
17. ionic bond
18. covalent bond
19. atom
20. molecule

SECTION 2. PROPERTIES OF WATER

1. false
2. true
3. true
4. a
5. b
6. c
7. evenly
8. solvent
9. nonpolar
10. more acidic, neutral, more basic
11. solute

SECTION 3. CARBON-BASED MOLECULES

1. true
2. false
3. true
4. false
5. Students should sketch one of the following, based on Figure 3.1 in the student text: straight chain, branched chain, or ring.
6. Provide energy
7. starches, sugar
8. Store energy
9. fat, oils
10. source of amino acids
11. beans, meat, nuts
12. map for making proteins
13. DNA, RNA
14. polymer

SECTION 4. CHEMICAL REACTIONS

1. reactants, products; reactants, products
2. chemical bonds
3. reactants
4. atoms
5. same rate
6. false
7. true
8. true
9. false
10. false
11. chemical reaction that absorbs more energy than it releases
12. chemical reaction that releases more energy than it absorbs
13. amount of energy that needs to be absorbed for a chemical reaction to start
14. substances changed during a chemical reaction
15. substances made by a chemical reaction
16. state reached when reactants and products are made at the same rate
17. amount of energy that will break a bond between two atoms

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JL Elias



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final action a reagent magnesium acetate solution dissolve 4.084 g mg ch₃coo h₂o in 0 ml h₂o and dilute to 1 l with alcohol b

determination from buret add 5 ml of the reagent to 3.5 g flour bread etc of 10 ml

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