

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
39
55
87
101
119

Atomic Mass	1.1
Atomic Number	1.2
Electron Configuration	1.3
Chemical Name	1.4
Chemical Symbol	1.5

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 16. It appears in the periodic table.
- _____ 17. It has no line periodic table.
- _____ 18. Group 1 elements.
- _____ 19. Elements that are shiny and conduct electricity.
- _____ 20. Group 8 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 21. There are two representative elements of elements.
- _____ 22. Most elements have lines in the periodic table for metalloids.
- _____ 23. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 24. Group 2A elements are the alkaline earth metals.
- _____ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 26. Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ 27. Elements have properties of both metals and nonmetals.

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