Economics suggestion (2nd Paper) For Dhaka Board Only H.S.C Examination—2015

Md. Abu Sayed Lecturer of Economics BAF Shaheen College, Dhaka Mobile: 01818374072

১--বাংলাদেশে অর্থনীতি পরিচয় ক) বাংলানেশের অর্থনীতির বৈশিষ্ট্যসমূহ আলোচনা কর। বীরে হলেও বাংলাদেশ ক্রমদ উল্লয়নের পথে ধাবিত হছেছ বিল্লেছণ কর। (বাংলাদেশের অর্থনীতির গতিধারা) গ) বাংলাদেশের অর্থনীতির ঐতিহাসিক পটভূমি বিপ্রেছণ কর। ক) বাংলাদেশের ভৌগলিক অবস্থান উল্লেখ কর। ব) অর্থনীতি কাঠামো কাকে বলে? ণ) সামাজিক ও অর্থনৈতিক অবকাঠামো কিং ২-বাংলাদেশের কৃষি ২। ক) বাংলাদেশের অর্থনৈতিক উন্নয়নে কৃষির ওকত্ব আলোচনা কর। বাংলাদেশের কৃষি পণ্য বিপশনের সমস্যা ও সমাধান। কৃষি ঋণ কি? বাংলাদেশের কৃষি ঋণের প্রাতিষ্ঠানিক ও অপ্রাষ্ট্রানিক উৎসদমূহ আলোচনা কর। থ) পরিবেশ দূষণ কিঃ বাংলাদেশের কৃষিতে পরিবেশ দূষণের কারণসমূহ আলোচনা কর। কিভাবে পরিবেশ দূষণ তথা জলবায়ু পরিবর্তনের প্রভাব মোকাবেলা করা যায়। शिका व ক) কৃষি জোত কি? ব্যাখ্যা কর। বৃষি পণ্য বিপণ্যবের সরকার বা রাষ্ট্রের ভূমিকা কত⊋কু। ণ) বৈশ্বিক উক্ষতা ও জলবায়ু পরিবর্তন বলতে কি বুঝা কৃষিতে তথ্য ও যোগাযোগ প্রযুক্তির (ICT) প্রযুক্তির ব্যবহার সম্পর্কে লিখ। কৃষিতে পরমাপু ও বাংলাটেকবোলজি পছতি কিঃ ছ) কৃষির বিভিন্ন উপথাততলো কিঃ ছ) কৃষির ভতুঁকি বলতে কি বুঝার? ৩-বাংলাদেশের শিস্ক ৪। ক) বাংলাদেশের তৈরি পোশাক শিল্পের বর্তমান সমস্যাবলি ও তবিষ্যত সম্ভাবনার বর্ণনা দাও। বাংলাদেশের আমলানি নির্ভর না হয়ে আমলানি বিকল্পন শিল্প গড়ে তোলা উচিতঃ তুমি কি একমতঃ গ) শিল্পোউন্নয়নে সরকারের গৃহীত পদক্ষেপসমূহ আলোচনা কর। ঘ) অর্থনৈতিক উন্নয়নে শিল্পের গুরুত্ব বিল্লেখন কর। ক) কুন্ত ও কুটির শিল্পের মধ্যে পার্থক্যঃ আমলানি বিকল্পন শিল্প বলতে কি বুঝা শ) মাঝারি শিল্পের বৈশিষ্ট্যওলো আলোচনা কর। ঘ) শিল্পনীতি-২০১০ এর মূল লক্ষ্য কিঃ ভ) শিল্পোয়ন্ত্ৰনে সরকারি-বেসরকারি অংশীদারিত (PPP) বলতে কি বুঝা চ) চামড়া শিল্পের অর্থনৈতিক গুরুত্ব বিশ্লেষণ কর। হ) বাংলাদেশের প্রধান প্রধান রক্তানি পশ্যকলো কি? ৪-জনসংখ্যা, মানবসম্পদ ও আত্রকর্মসংস্থান ৪ । ক) বাংলাদেশের অর্থনীতিতে জনসংখ্যা বৃদ্ধির প্রভাব বিশ্রেষণ কর । খ) ফ্রালখালের জনসংখ্যা তত্ত্বতি সমালোচনাসহ ব্যাখ্যা কর। গ) বাংলাদেশের জনসংখ্যার বৈশিষ্ট্যসমূহ আলোচনা কর। ঘ) মানবসম্পদ উন্নয়ন কিঃ মানবসম্পন উন্নয়নের সরকারি গৃহীত পদক্ষেপসমূহ কিঃ शिका इ ক) বেকারত কি? খ) বেকারত্বের শ্রেণি বিভাগ লিখ। ণ) একটি বাড়ি একটি খামার প্রকল্প কিঃ তেকার সমস্যা সমাধানের সরকারের বিভিন্ন শদক্ষেপ লিব। জনসংখ্যা ও বেকারত্বের সম্পর্ক শিব। চ) আত্রকর্মসংস্থান বলতে কি বৃঝঃ কাম্য জনসংখ্যা ধারণাটি ব্যাখ্যা কর। ক) বাংলাদেশের জনসংখ্যা দ্রুত বৃদ্ধির কারণ লিখ। ৫-খাদ্য নিরাপতা ৫। ক) বাংলাদেশে বর্তমান খাল্য নিরাপদ্যা পরিস্থিতি ব্যাখ্যা কর। নিরাপন খাদ্য কিঃ নিরাপন খাদ্যের গুরুত্ব ব্যাখ্যা কর।

বাকী পৃষ্ঠাগুলি নিচের লিচ্ছে ক্লিক করে ডাউনলোড করে পড়ুন।

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web the purpose of the copper iron stoichiometry lab was to determine which of the two possible iron sulfate compounds formed when reacting aqueous copper ii sulfate with solid iron metal the techniques used in this lab were weighing by difference quantitative transfer and vacuum filtration

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate cuso4 and 2 0153 grams of iron powder fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared stoichiometry the reaction of iron with copper ii sulfate - Jun 13 2023

web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely

divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution ofcopper ii sulfate *lab report 4 stoichiometry of iron copper ii sulfate -* Feb 09 2023

web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated cuso4 solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g