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2011 to 2014 (60 to 64 reg) Petrol & Diesel

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step-by-step maintenance and repair



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web 1 a 4 31 x 10<sup>22</sup> atoms ni b 7 01 x 10<sup>20</sup> ions br c 2 52 x 10<sup>23</sup> molecules h 2 d 7 79 x 10<sup>21</sup> molecules h 2 o 2 2 a 5 25 g naoh b 1 15 x 10<sup>3</sup> g lactose c 82 1 g fe d 0 332 g no 2 3 6 54 x 10<sup>21</sup> molecules of c 3 h 8 o 3 4 2 79 grams of nh 3 5 1 97 x 10<sup>22</sup> grams of sn

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web chapter 10 chemical quantities basics the basic unit that is used to determine the amount of a chemical substance is called a mole a mole mol of a substance is equivalent to 6 02 x 10<sup>23</sup> particles of that substance

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