

1. Known: Atomic number of Ru is 44.

Unknown: Electron configuration of Ru (ruthenium).

Nearest noble gas is Kr with atomic number 36. Ground-state electron configuration is  $[\text{Kr}]4d^75s^1$ .

2.

1. 44

2. 44

3.  $[\text{Kr}]4d^75s^1$

4. 8

5. 8

6. 4d and 5s

7.  $[\text{Kr}]4d^75s^1$  or  $1s^22s^22p^63s^23p^64s^23d^{10}4p^65s^14d^7$

3. 44, Kr, Aufbau principle

# Glencoe Science Notebook Chemistry Answers Chapter 7

**AW Rasmussen**



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