

- 8) a) $2.96 - 0.82 = 2.14$
b) (MOSTLY) IONIC
c) rubidium bromide
d) BI
- 9) a) $3.16 - 2.55 = 0.61$
b) POLAR COVALENT
c) carbon tetrachloride
d) BM
- 10) a) $2.55 - 1.65 = 0.90$
b) POLAR COVALENT
c) zinc acetate
d) other
- 11) a) $3.98 - .93 = 3.05$
b) (MOSTLY) IONIC
c) sodium fluoride
d) BI
- 12) a) $3.44 - 3.44 = 0$
b) NONPOLAR COVALENT
c) oxygen
d) BM
- 13) a) $3.04 - 1.90 = 1.14$
b) POLAR COVALENT
c) copper(I) nitride
d) BI
- 14) a) $3.16 - 0.82 = 2.34$
b) (MOSTLY) IONIC
c) potassium chlorate
d) TI
- 15) a) $2.58 - 0.95 = 1.63$
b) POLAR COVALENT
c) strontium sulfide
d) BI

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M Lipman



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