

# Chemistry and Matter Section 1.2 and 1.3. Read these sections first.

Define each term.

1. chemistry

2. matter

3. mass

Write each term below under the correct heading. Use each term only once.

air	magnetic field	car	feeling	heat	human body
light	radio	radio wave	flashlight	textbook	thought
Made of Matter				Not Made of Matter	
4.					16.
5.					17.
6.					18.
7.					19.
8.					20.
9.					21.

For each statement below, write *true* or *false*.

16. The mass of an object can vary with the object's location.

17. A mass measurement includes the effect of Earth's gravitational pull on the object being measured.

18. Scientists measure the amount of matter in terms of mass.

19. Subtle differences in weight exist at different locations on Earth.

20. Your mass on the Moon would be smaller than your mass on Earth.

Circle the letter of the choice that best completes the statement or answers the question.

1. Which of the following is **NOT** matter? a. stone b. ultraviolet radiation c. air d. the Sun

2. At the end of an experiment, the scientist forms a conclusion based on the a. variable. b. scientific law. c. data obtained. d. control.

3. Which of the following is a set of controlled observations that tests a hypothesis?

a. mass b. experiment c. weight d. constant

4. The branch of chemistry that focuses on carbon-containing chemicals is called a. analytical chemistry. b. inorganic chemistry. c. biochemistry. d. organic chemistry.

5. How should you prepare an acid solution?

a. Add the water to the acid all at once. c. Add the water to the acid very slowly.

b. Add the acid to the water all at once. d. Add the acid to the water very slowly.

Identify each piece of data as either *qualitative* or *quantitative*.

1. red

2. 100 pounds

3. 105°C

4. tall

5. round

6. smells like bananas

7. 40 mph

8. pink with purple polka dots

9. cold

10. 70 books

Identify each kind of investigation as an example of *pure research* or *applied research*.

# Glencoe Chemistry Matter Change Answer Key Chapter 1

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