

4. **Think and discuss:** Why is there an attraction between the two ions in this chemical bond?

There is an attraction between these two ions in this chemical bond due to the fact that there is a positively charged atom (cation) and a negatively charged atom (anion). Opposite forces attract (+, -) and same forces repel (+, + and -, -).

Activity B: Ionic compounds	Get the Game ready: <ul style="list-style-type: none"> Click Reset. Turn off Show charge. Select Lithium (Li) and Oxygen (O). 	
---------------------------------------	--	---

Question: How are ionic compounds formed?

1. **Observe:** Look at the red lithium atom and the blue oxygen atom. Recall that most atoms are stable when their outermost ring has eight electrons. (Some atoms, such as lithium and beryllium, are stable when their outermost ring has two electrons.)

- A. How many electrons will the lithium atom give up to become stable?
- B. How many electrons does the oxygen atom need to become stable?
- C. Can a stable compound be made from these two atoms? Explain why or why not.

A stable compound cannot be made from these two atoms. Oxygen needs 2 electrons to have a stable orbit, and be happy. Though Lithium cannot give up 2 electrons without being unhappy due to the fact that it would not have a stable orbit.

2. **Experimentally:** Click **Add metal** to add another lithium atom, and then transfer electrons from the lithium to the oxygen. Click **Check**.

- A. Did you make a stable compound?

- B. Turn on **Show formula**. What is the formula of this compound?

- C. Turn on **Show charge**. What is the charge of each ion? Li O

3. **Design:** Use the Game to create stable compounds from the combinations given below. After transferring electrons, arrange the atoms to demonstrate the attraction between positively charged ions and negatively charged ions. Click **Check** to check each compound.

For each compound, click the **camera** (📷) icon to take a snapshot. Right-click the image, and click **Copy Image**. Paste each image into a blank document to turn in with this worksheet. Write the ionic charges (such as Ca^{2+}) and chemical formulas below.

	Ionic charges	Chemical formula
A. Lithium and fluorine:	Li <input type="text" value="+1"/> F <input type="text" value="-1"/>	<input type="text" value="LiF"/>

Gizmos Answers Ionic Bonds

Michal Rosen-Zvi

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Gizmos Answers Ionic Bonds:

Reviewing **Gizmos Answers Ionic Bonds**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Gizmos Answers Ionic Bonds**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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