



Name: Tra Nguyen

Date: 11/12/2020

Student Exploration: Electron Configuration

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: atomic number, atomic radius, Aufbau principle, chemical family, diagonal rule, electron configuration, Hund's rule, orbital, Pauli exclusion principle, period, shell, spin, subshell

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Elvis Perkins, a rather shy fellow, is getting on the bus shown at right. Which seat do you think he will probably sit in? Mark this seat with an "E."

Elvis Perkins will most likely sit on the seat on the upper left that is empty with no passengers.

2. Marta Warren gets on the bus after Elvis. She is tired after a long day at work. Where do you think she will sit? Mark this seat with an "M."

Marta Warren will most likely sit next to the door with the white and black fellow. Since she is tired and will want to not have to walk all the way down the rows when it is her time to exit.

3. In your experience, do strangers getting on a bus like to sit with other people if there is an empty seat available? _____

No, people tend to sit and give others space if there is an empty seat available on a bus.



Gizmo Warm-up

Just like passengers getting on a bus, electrons orbit the nuclei of atoms in particular patterns. You will discover these patterns (and how electrons sometimes act like passengers boarding a bus) with the *Electron Configuration Gizmo*.

To begin, check that **Lithium** is selected on the PERIODIC TABLE tab.

1. The **atomic number** is equal to the number of protons in an atom.

How many protons are in a lithium atom? _____

There are 3 protons in a Lithium atom.



3	4	5	6	7	8
Sc	Ti	V	Cr	Mn	Fe
Y	Zr	Nb	Mo	Tc	Ru
Lr	Hf	Ta	W	Re	Os

2. A neutral atom has the same number of electrons and protons.
How many electrons are in a neutral lithium atom? _____

Gizmo Student Exploration Electron Configuration Answers

G Thomas

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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