

## Gizmo #1: Student Exploration: Nuclear Decay

**Directions:** Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

**Vocabulary:** alpha particle, atomic number, beta particle, daughter product, gamma ray, isotope, mass number, nuclear decay, positron, radioactive, subatomic particle

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

The chart below gives the locations, charges, and approximate masses of three **subatomic particles**. The approximate mass of each particle is given in universal mass units (u).

Particle	Location	Charge	Approximate mass
Proton	Nucleus	$1^+$	1 u
Neutron	Nucleus	0	1 u
Electron	Orbitals	$1^-$	0 u

- The **mass number** of an atom is equal to the sum of protons and neutrons in the nucleus. A helium atom has 2 protons and 2 neutrons. What is the mass number of this atom?

4

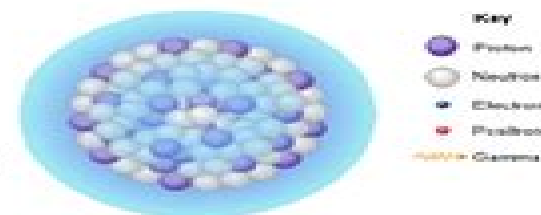
- The **atomic number** of an element is equal to the number of protons in each atom of the element. All helium atoms have 2 protons. What is the atomic number of helium?

2

### Gizmo Warm-up

While most atoms are stable, some are **radioactive**, which means that they have a tendency to undergo spontaneous **nuclear decay**. The decay of radioactive atoms generally results in the emission of particles and/or energy.

Several types of nuclear decay can be explored with the *Nuclear Decay* Gizmo. On the Gizmo, check that **Alpha decay** and **Uranium** are selected.



- Click **Play** (▶), and then click **Pause** (⏸) when the **alpha particle** is clearly visible. What is an alpha particle made of? Protons and neutrons
- Click **Play** and observe. Besides the alpha particle, what else is emitted from the nucleus during alpha decay? Gamma

**Gamma rays** are energetic electromagnetic waves; they are often emitted in nuclear decay.

# Gizmo Nuclear Decay Answers

**Huangqi Zhang**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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## **Table of Contents Gizmo Nuclear Decay Answers**

1. Understanding the eBook Gizmo Nuclear Decay Answers
  - The Rise of Digital Reading Gizmo Nuclear Decay Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Nuclear Decay Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gizmo Nuclear Decay Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Nuclear Decay Answers

- Personalized Recommendations
  - Gizmo Nuclear Decay Answers User Reviews and Ratings
  - Gizmo Nuclear Decay Answers and Bestseller Lists
5. Accessing Gizmo Nuclear Decay Answers Free and Paid eBooks
    - Gizmo Nuclear Decay Answers Public Domain eBooks
    - Gizmo Nuclear Decay Answers eBook Subscription Services
    - Gizmo Nuclear Decay Answers Budget-Friendly Options
  6. Navigating Gizmo Nuclear Decay Answers eBook Formats
    - ePub, PDF, MOBI, and More
    - Gizmo Nuclear Decay Answers Compatibility with Devices
    - Gizmo Nuclear Decay Answers Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Gizmo Nuclear Decay Answers
    - Highlighting and Note-Taking Gizmo Nuclear Decay Answers
    - Interactive Elements Gizmo Nuclear Decay Answers
  8. Staying Engaged with Gizmo Nuclear Decay Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Gizmo Nuclear Decay Answers
  9. Balancing eBooks and Physical Books Gizmo Nuclear Decay Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Gizmo Nuclear Decay Answers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Gizmo Nuclear Decay Answers
    - Setting Reading Goals Gizmo Nuclear Decay Answers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Gizmo Nuclear Decay Answers

- 
- Fact-Checking eBook Content of Gizmo Nuclear Decay Answers
  - Distinguishing Credible Sources

### 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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