

## PRACTICAL ORGANIC CHEMISTRY

### (A) FUNCTIONAL GROUP ANALYSIS

1. **Unsaturation : Alkenes & alkynes:**
    - (a) Bayers test : Cold dil alk.  $\text{KMnO}_4$  decolourisation test  
Purple colour  $\rightarrow$  Colourless +  $\text{MnO}_2$  (Brown ppt)
    - (b)  $\text{Br}_2$  water decolourisation test  
Violet colourless of  $\text{Br}_2 \rightarrow$  Colourless
  2. **Terminal alkynes:**  
Confirmed by ppt of Acetylide ion with  $\text{NaNH}_2$  or  $\text{AgNO}_3$  or  $\text{Cu}_2\text{Cl}_2\text{NH}_4\text{OH}$
  3. **Alkyl halides:**
    - (a) If they are capable of carbocation formation then they will give ppt with  $\text{AgNO}_3$ .
    - (b) Beilstein's test : A green colour is imported to the flame if small amount of organic compound is taken on copper wire.
  4. **Alcohol:**
    - (a) Ferric ammonium nitrate  $\rightarrow$  Give red colour
    - (b) Boil with acetic acid & conc.  $\text{H}_2\text{SO}_4 \rightarrow$  fruity smell
    - (c) 2-alkanol & ethanol also give Iodoform test  $\rightarrow$  Yellow ppt. of  $\text{CHI}_3$  on reaction with  $\text{I}_2 + \text{OH}^-$
  5. **Aldehyde & Ketones:**  
2,4- Dinitrophenyl hydrazine (or) Braddy's reagent give yellow, orange or red color with ald. & Ketones (2,4-DNP)
  6. **Aldehydes:**
    - (a) Tollen's test  $\rightarrow$  Silver mirror
    - (b) Fehling's test [ except benzaldehyde ]  $\rightarrow$  Red colour
    - (c) Benedicts test  $\rightarrow$  Red colour
    - (d) Schiff's dye colour regeneration test  $\rightarrow$  Pink colour
    - (e) Gly ppt with  $\text{HgCl}_2$ .
  7. **Ketones:**
    - (a) Methyl Ketones give haloform test
    - (b)  $\alpha$ -hydroxy Ketones give Tollen's & Fehling test's too.
  8. **Carboxylic acids:**
    - (a) Brisk effervescence with aq.  $\text{NaHCO}_3$  solution.
    - (b)  $\text{HCOOH}$  alone gives silver mirror test with Tollen's reagent.
    - (c) Blue litmus  $\rightarrow$  red
    - (d) Give fruity smell on reaction with alcohols.
  9. **Phenols:**
    - (a) Violet colouration with neutral  $\text{FeCl}_3$
    - (b) Liebermann test
    - (c) White ppt with  $\text{Br}_2$  water
    - (d) Brisk effervescence with aq.  $\text{NaHCO}_3$  is observed in case of Nitrophenols.
  10. **Primary amines:**
    - (a) Carbylamine reaction  $\rightarrow$  Isonitriles have very distinctive foul odors
    - (b) Hoffmann mustard oil reaction  $\rightarrow$  Oily liquid with mustard like smell.
  11. **Aromatic 1° amine  $\rightarrow$  diazo test**
-

# Give Me D Answer Of Chemistry Pratical Alternative A

**Patrick Vollmar**



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