
Quiz

Section: Exploring the Periodic Table

In the space provided, write the letter of the element whose most likely properties match a characteristic listed in each paragraph.

- _____ 1. Belong to the families of elements in the periodic table represented as follows:

a. metal	c. nonmetal halogen
b. member of actinides	d. transition metal
- _____ 2. Characterized largely by the tendency to lose the outer electrons of a valence electron:

a. noble gas	c. metal halogen
b. period elements	d. nonmetals
- _____ 3. Because the pattern is similar to groups, noble gases and the transition elements have an filled outer energy level, elements in a group:

a. have different charges	c. are very reactive
b. share nearly the same	d. are pure
- _____ 4. A group that acts as low transition metal:

a. metals	c. noble
b. metalloids	d. halogens
- _____ 5. Characterized by low density, low melting point, and low boiling point:

a. halogen halogen	c. noble halogen
b. noble gas	d. transition metal
- _____ 6. Group of elements that:

a. metals	c. metalloids
b. noble	d. nonmetals
- _____ 7. The three main groups of elements are metals, metalloids, and:

a. noble gases	c. noble gas halogens
b. noble metalloids	d. nonmetals
- _____ 8. Elements that:

a. metals	c. metalloids
b. metalloids	d. nonmetals
- _____ 9. Elements that have the full outer shell of the periodic table and:

a. noble metalloids	c. metalloids
b. noble	d. nonmetals
- _____ 10. Elements that:

a. noble	c. metalloids
b. noble metalloids	d. noble

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Ying Liu



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