Name: Date:

Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

If Nate were to give Clara one of his markers, then both would end up with eight.

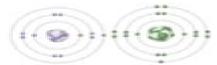
Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

If Maggie gives Fred and Florence one marker each, they will all have eight markers.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *lanic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that Sodium (Na) and Chlorine (CI) are selected from the menus at right. Click Play (- to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



 Each atom consists of a central nucleus and several shells that contain electrons. The outermost electrons are called valence electrons.

How many valence electrons does each atom have? Sodium: 1 Chlorine:

7

Click Pause ("). Elements can be classified as metals and nonmetals. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. Electron affinity is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

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AW Chickering

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Table of Contents Explore Learning Gizmo Activity Answers Ionic Bonds

- 1. Understanding the eBook Explore Learning Gizmo Activity Answers Ionic Bonds
 - The Rise of Digital Reading Explore Learning Gizmo Activity Answers Ionic Bonds
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Explore Learning Gizmo Activity Answers Ionic Bonds
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Gizmo Activity Answers Ionic Bonds
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Explore Learning Gizmo Activity Answers Ionic Bonds
 - Personalized Recommendations
 - Explore Learning Gizmo Activity Answers Ionic Bonds User Reviews and Ratings
 - Explore Learning Gizmo Activity Answers Ionic Bonds and Bestseller Lists
- 5. Accessing Explore Learning Gizmo Activity Answers Ionic Bonds Free and Paid eBooks
 - Explore Learning Gizmo Activity Answers Ionic Bonds Public Domain eBooks
 - Explore Learning Gizmo Activity Answers Ionic Bonds eBook Subscription Services
 - Explore Learning Gizmo Activity Answers Ionic Bonds Budget-Friendly Options

- 6. Navigating Explore Learning Gizmo Activity Answers Ionic Bonds eBook Formats
 - o ePub, PDF, MOBI, and More
 - Explore Learning Gizmo Activity Answers Ionic Bonds Compatibility with Devices
 - Explore Learning Gizmo Activity Answers Ionic Bonds Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Gizmo Activity Answers Ionic Bonds
 - Highlighting and Note-Taking Explore Learning Gizmo Activity Answers Ionic Bonds
 - o Interactive Elements Explore Learning Gizmo Activity Answers Ionic Bonds
- 8. Staying Engaged with Explore Learning Gizmo Activity Answers Ionic Bonds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Gizmo Activity Answers Ionic Bonds
- 9. Balancing eBooks and Physical Books Explore Learning Gizmo Activity Answers Ionic Bonds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Gizmo Activity Answers Ionic Bonds
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Explore Learning Gizmo Activity Answers Ionic Bonds
 - Setting Reading Goals Explore Learning Gizmo Activity Answers Ionic Bonds
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Explore Learning Gizmo Activity Answers Ionic Bonds
 - Fact-Checking eBook Content of Explore Learning Gizmo Activity Answers Ionic Bonds
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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