

Chapter 9 Carbon Chemistry

Section 9.2 Substituted Hydrocarbons**(pages 272–274)**

This section discusses organic compounds that contain atoms of elements other than carbon and hydrogen. It also explains the relationship between the properties of organic compounds and functional groups.

Reading Strategy (page 272)

Monitoring Your Understanding As you read, complete the table by connecting each functional group with the type of compound that contains the functional group. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Connecting Functional Groups to Types of Compounds	
Functional Group	Type of Compound
–OH	
–COOH	
–NH ₂	

- Name the two main products when methane and chlorine react.
 - _____
 - _____
- To which environmental problem have researchers connected halocarbons containing chlorine and fluorine? _____

- Describe a substituted hydrocarbon. _____

- Is the following sentence true or false? The functional group in a substituted hydrocarbon determines the properties of the compound. _____

Alcohols (page 273)

- Methanol and ethanol are two examples of a class of organic compounds called _____.
- The functional group in an alcohol is represented as –OH and is called a(n) _____ group.
- Identify two ways a halocarbon can be produced.
 - _____
 - _____

Hydrocarbon Section Review Answers

**Barron's Educational Series, Mark
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Table of Contents Hydrocarbon Section Review Answers

1. Understanding the eBook Hydrocarbon Section Review Answers
 - The Rise of Digital Reading Hydrocarbon Section Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Hydrocarbon Section Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hydrocarbon Section Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hydrocarbon Section Review Answers
 - Personalized Recommendations
 - Hydrocarbon Section Review Answers User Reviews and Ratings
 - Hydrocarbon Section Review Answers and Bestseller Lists
5. Accessing Hydrocarbon Section Review Answers Free and Paid eBooks
 - Hydrocarbon Section Review Answers Public Domain eBooks
 - Hydrocarbon Section Review Answers eBook Subscription Services
 - Hydrocarbon Section Review Answers Budget-Friendly Options
6. Navigating Hydrocarbon Section Review Answers eBook Formats

- ePub, PDF, MOBI, and More
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- Hydrocarbon Section Review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hydrocarbon Section Review Answers
 - Highlighting and Note-Taking Hydrocarbon Section Review Answers
 - Interactive Elements Hydrocarbon Section Review Answers
- 8. Staying Engaged with Hydrocarbon Section Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hydrocarbon Section Review Answers
- 9. Balancing eBooks and Physical Books Hydrocarbon Section Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hydrocarbon Section Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Hydrocarbon Section Review Answers
 - Setting Reading Goals Hydrocarbon Section Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hydrocarbon Section Review Answers
 - Fact-Checking eBook Content of Hydrocarbon Section Review Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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web timing control valve is duty controlled according to a signal from ecm engine control module to control the fuel injection start timing using the crankshaft angle feed back system highly precise control is effected

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