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2. Next, what type of bond is connecting the two amino acids? covalent bond or peptide bond. If it is covalent, then you will have CO, C=O, NH, and OH on each amino acid. If it is ionic, then you will have COO⁻ and NH⁺ on each amino acid. If it is hydrogen, then you will have CO and NH on each amino acid. If it is disulfide, then you will have two SH groups on each amino acid.

3. Now, what type of bond is connecting the two amino acids? covalent bond or peptide bond. If it is covalent, then you will have CO, C=O, NH, and OH on each amino acid. If it is ionic, then you will have COO⁻ and NH⁺ on each amino acid. If it is hydrogen, then you will have CO and NH on each amino acid. If it is disulfide, then you will have two SH groups on each amino acid.

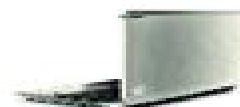
4. Finally, what type of bond is connecting the two amino acids? covalent bond or peptide bond. If it is covalent, then you will have CO, C=O, NH, and OH on each amino acid. If it is ionic, then you will have COO⁻ and NH⁺ on each amino acid. If it is hydrogen, then you will have CO and NH on each amino acid. If it is disulfide, then you will have two SH groups on each amino acid.

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