

## CHAPTER 21 REVIEW

***Nuclear Chemistry*****SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. \_\_\_\_\_ Based on the information about the three elementary particles in the text, which has the greatest mass?
  - (a) the proton
  - (b) the neutron
  - (c) the electron
  - (d) They all have the same mass.
2. \_\_\_\_\_ The force that keeps nucleons together is
  - (a) a strong nuclear force.
  - (b) a weak nuclear force.
  - (c) an electromagnetic force.
  - (d) a gravitational force.
3. \_\_\_\_\_ The stability of a nucleus is most affected by the
  - (a) number of neutrons.
  - (b) number of protons.
  - (c) number of electrons.
  - (d) ratio of neutrons to protons.
4. \_\_\_\_\_ If an atom should form from its constituent particles,
  - (a) matter is lost and energy is taken in.
  - (b) matter is lost and energy is released.
  - (c) matter is gained and energy is taken in.
  - (d) matter is gained and energy is released.
5. \_\_\_\_\_ For atoms of a given mass number, those with greater mass defects, have
  - (a) smaller binding energies per nucleon.
  - (b) greater binding energies per nucleon.
  - (c) the same binding energies per nucleon as those with smaller mass defects.
  - (d) variable binding energies per nucleon.
6. Based on **Figure 1.1** of the text, which isotope of He, helium-3 or helium-4,
  - \_\_\_\_\_ a. has the smaller binding energy per nucleon?
  - \_\_\_\_\_ b. is more stable to nuclear changes?
7. The number of neutrons in an atom of magnesium-25 is \_\_\_\_\_.
8. Nuclides of the same element have the same \_\_\_\_\_.

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*Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook* McGraw Hill, 2001-08  
*Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change* **Chemistry** John S. Phillips, Cheryl Wistrom, 2000  
*Chemistry* McGraw-Hill Staff, 2001-03 Glencoe Mathematics William Collins, 1999 **Solutions Manual to Accompany Elements of Organic Chemistry** Isaak Zimmerman, Henry Zimmerman, 1977 **Astronomy** Alton Biggs, 2002  
**Science Interactions**, 1996 Teen Health Course 3 Mary Helen Bronson, Glencoe/McGraw-Hill, 1999 **Glencoe Physical iScience, Grade 8, Reading Essentials, Student Edition** McGraw Hill, 2004-08-02  
Reading Essentials student edition provides an interactive reading experience to improve student comprehension of science content. It makes lesson content more accessible to struggling students and supports goals for differentiated instruction. Students can highlight text and take notes right in the book.  
The Dynamics of Science and Technology W. Krohn, E.T. Layton Jr., P. Weingart, 2012-12-06  
The interrelations of science and technology as an object of study seem to have drawn the attention of a number of disciplines: the history of both science and technology, sociology, economics, and economic history, and even the philosophy of science. The question that comes to mind is whether the phenomenon itself is new or if advances in the disciplines involved account for this novel interest or in fact if both are interconnected. When the editors set out to plan this volume, their more or less explicit conviction was that the relationship of science and technology did reveal a new configuration and that the disciplines concerned with its analysis failed at least in part to deal with the change because of conceptual and methodological preconceptions. To say this does not imply a verdict on the insufficiency of one and the superiority of any other disciplinary approach. Rather, the situation is much more complex. In economics, for example, the interest in the relationship between science and technology is deeply influenced by the theoretical problem of accounting for the factors of economic growth. The primary concern is with technology and the problem is whether the market induces technological advances or whether they induce new demands that explain the subsequent diffusion of new technologies. Science is generally considered to be an exogenous factor, not directly subject to market forces and therefore appears to be of no interest.  
Merrill Chemistry Robert C. Smoot, Richard G. Smith, Jack Price, 1998 Enrichment Chemistry Modern Course GLENCOE, 1995-04 Merrill Earth Science Ralph M. Feather, Susan Leach Snyder, Dale T. Hesser, 1995  
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