Name:

Ashley Maddison

Date:

11/28/2020

### Student Exploration: Covalent Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers?

Frank should give the red marker to Flora, and Flora should give the blue marker to Frank.

Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Otto should give the black and brown makers to Olivia, and Olivia should give the purple and green markers to Otto.

#### Gizmo Warm-up

Just like the students described above, **norm etal** atoms can share electrons As you will see in the Covalent Bonds Gizmo, atoms form bonds in this way.

To begin, check that Fluorine is selected from the Select a substance menu. Click Play ( E) to see the electrons orbiting the nucleus of each atom.



The outermost electrons in each atom are called valence electrons. How many valence electrons does

each fluorine atom have? 7 valence electrons.

2. Click Pause ( !!! ). Drag an electron from the left atom to the right atom. Click Play.

What happens? They equally share the one atom on the new shell that is formed.

3. Click Pause, drag an electron from the right atom to the left, and then click Play.

# **Gizmo Student Exploration Covalent Bonds Answers**

O García

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