

### Activity A:

#### Subatomic particles

#### Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



**Question:** What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** ( $A$ ), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** ( $Z$ ).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons ( $N$ ) in an atom?
- C. Which particle (proton, neutron, or electron) has a  
Positive charge?  
Negative charge?

neutrons

by the number of protons

electron

proton

neutron

3. **Explore:** Use the Gizmo to create different combinations of particles. Each combination is an atom, but some combinations are not. Record the number of protons, neutrons, and electrons in each atom. The number is represented by the element symbol and the mass number. What do you notice? Some combinations result in atoms, while others are **ions**, which means the atom is charged and does not exist naturally.

- A. What are the stable isotopes of hydrogen?
- B. What are the stable isotopes of helium?
- C. Carbon has three isotopes of oxygen.

1 proton and 0 neutrons

2 protons and 2 neutrons

2

4. **Challenge:** Use the Gizmo to answer the following questions.

- A. How many electrons are in a neutral atom of lithium?
- B. How many electrons are in an atom of hydrogen?
- C. What is the mass number of an atom with 3 protons and 3 neutrons?
- D. What is an atom's charge if it contains 11 protons and 10 electrons?
- E. How many electrons are in hydrogen?

3

1

6

1

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# Gizmo Element Builder Answer Key

**SA Dillow**



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