



Student Exploration: Stoichiometry

Background: A chemical equation is a shorthand representation of a chemical reaction. It shows the chemical formulas of the reactants and products, their relative amounts, and the conditions under which the reaction takes place.

Activity 1:
Stoichiometry

Read the problem carefully.

1. Consider the equation below:



Problem

Background: Iron(III) oxide (hematite ore) for iron is reduced (iron metal and carbon monoxide) using the following chemical equation: $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$. This process is an example of a chemical reaction. A chemical reaction is a process in which one or more substances are converted into one or more new substances. The substances that enter the reaction are called **reactants**, and the substances that are produced are called **products**. The equation above shows that iron(III) oxide and carbon monoxide are the reactants, and iron metal and carbon dioxide are the products.



Chemical equations are balanced. This means that the number of atoms of each element is the same on both sides of the equation. In the equation above, there are 2 iron atoms and 3 carbon atoms on both sides. This is why the equation is balanced.

Question: What is the ratio of the coefficients?



- What is the ratio of the coefficients of Fe_2O_3 and Fe ?
- What is the ratio of the coefficients of Fe_2O_3 and CO_2 ?
- What is the ratio of the coefficients of Fe_2O_3 and CO ?
- What is the ratio of the coefficients of Fe and CO_2 ?

Question: How many atoms of each element are there on each side?

- Calculate the total number of atoms of each element on each side of the equation. $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$. How many atoms of each element are there on each side?



Explore Learning Gizmo Answer Key Stoichiometry

Didier Musso



Explore Learning Gizmo Answer Key Stoichiometry:

The Enigmatic Realm of **Explore Learning Gizmo Answer Key Stoichiometry**: Unleashing the Language is Inner Magic

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