6.	Observe: Click Reset. Move the Reactant concentration slider back and forth.	What do
	you notice?	

You soo mare & lease reasons & products appear

7. Predict: How will increasing the reactant concentration affect the rate of the reaction? Why?

The note readonts the more products will be preated

Gather data: Make sure the Temperature is 200 °C and the Surface area is Maximum.
Use the Gizmo to measure the half-life for each given reactant concentration. (Note that the number of reactant molecules changes with each concentration.) Calculate the means.

Trial	0.4 mol/L	0.8 mol/L	1,2 mol/L	1.6 mol/L	2.0 mol/L
1	3:52	4:18	1143	1:02	1:08
2	4:63	1:13	1:02	0156	0138
Mean half-life	3,77	2,604	1,22	-76	.73

Compare: If possible, find the mean times for each concentration for your entire class. What
is the mean class time for a concentration of 0.4 mol/L? How about for 2.0 mol/L?

Mean for 0.4 mol/L: ____3.77

1.01021

10. Analyze: What do these results indicate? Sill the 2.0 moi/L is the

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M Lipman

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