# Worlds highest boundaries incoming? AQA A-Level Chemistry Paper 2 Unofficial markscheme:

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The total mark for this paper was 105 with 9 questions. Please add to it as much you can. The above is the archive, however this document can still be edited for debate/additional information. Please keep in mind this is a student made mark scheme, and these answers may not all be what is on the official mark scheme.

#### Topic for Paper 3: DNA, Chromatography, aspirin production

Question	Answer	Additional Information/Debate	Mark
1 why use a conical flask instead of a beaker?	I put it has a smaller opening than a beaker so less gas lost?? You want to lose gas? - I am pretty sure this is the right answer, well it decreases the rate of gas no loss so easier to measure the rate. It doesn't reduce the rate of gas loss as this only depends on the concentration of reactants  No splashing This was jokes Reactants don't stick to the side what?????? they would stick a little regardless	what was the cotton wool in that diagram for lotz it threw me off.  I put that the liquid would be shallower with larger surface areaso gas produced could escape quicker.  I put larger surface area for to heat mixture more evenly? I said this too.  You don't want to have a large opening be the acid may be lost due to the bubbles forming.  Anything about swirling? I wrote swirling	an:
rate at 2 minutes.	0.5 g min-1? This is correct. A range of values typically accepted. I got 0.4 so yeah they'll accept a range AS LONG as they can see how you worked it out.  G min-1? Yeah but if u changed to seconds it wouldn't really matter as long as u put g per second  Mol Dm-3s-1 +	I got 0.02, you had to convert to mols - you didn't have to, any appropriate units would be fine I got like 0.0175 gs-1 - Same here - yeah just rounded Gmin-1 no? Is it not gmin-1 the graph was in minutes I converted the minutes to seconds and put my units with seconds as well, any idea if thats okay? Yes	3

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#### **Table of Contents M1 Unofficial Mark Scheme 2015**

- 1. Understanding the eBook M1 Unofficial Mark Scheme 2015
  - The Rise of Digital Reading M1 Unofficial Mark Scheme 2015
  - Advantages of eBooks Over Traditional Books
- 2. Identifying M1 Unofficial Mark Scheme 2015
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an M1 Unofficial Mark Scheme 2015
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from M1 Unofficial Mark Scheme 2015

- Personalized Recommendations
- M1 Unofficial Mark Scheme 2015 User Reviews and Ratings
- M1 Unofficial Mark Scheme 2015 and Bestseller Lists
- 5. Accessing M1 Unofficial Mark Scheme 2015 Free and Paid eBooks
  - M1 Unofficial Mark Scheme 2015 Public Domain eBooks
  - M1 Unofficial Mark Scheme 2015 eBook Subscription Services
  - M1 Unofficial Mark Scheme 2015 Budget-Friendly Options
- 6. Navigating M1 Unofficial Mark Scheme 2015 eBook Formats
  - o ePub, PDF, MOBI, and More
  - M1 Unofficial Mark Scheme 2015 Compatibility with Devices
  - M1 Unofficial Mark Scheme 2015 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of M1 Unofficial Mark Scheme 2015
  - Highlighting and Note-Taking M1 Unofficial Mark Scheme 2015
  - Interactive Elements M1 Unofficial Mark Scheme 2015
- 8. Staying Engaged with M1 Unofficial Mark Scheme 2015
  - o Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers M1 Unofficial Mark Scheme 2015
- 9. Balancing eBooks and Physical Books M1 Unofficial Mark Scheme 2015
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection M1 Unofficial Mark Scheme 2015
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine M1 Unofficial Mark Scheme 2015
  - Setting Reading Goals M1 Unofficial Mark Scheme 2015
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of M1 Unofficial Mark Scheme 2015

- Fact-Checking eBook Content of M1 Unofficial Mark Scheme 2015
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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