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ভারতশিখারি: বিজ্ঞানই হল অস্ত্রসংগ্রহস্থল

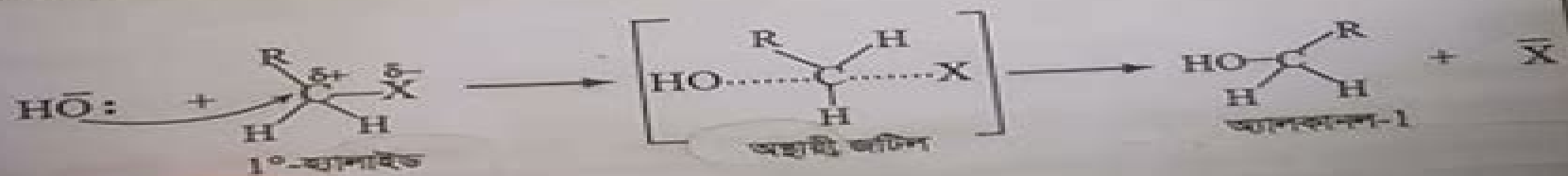
S_N1 বিক্রিয়ার ক্ষেত্রে RX এর সক্রিয়তার ক্রম হলো : $3^\circ RX > 2^\circ RX > 1^\circ RX$

S_N2 বা দ্বি-আণবিক নিউক্লিওফিলিক প্রতিস্থাপন : যে S_N বিক্রিয়ার হার বা গতি নিউক্লিওফাইল ও অ্যালকাইল হ্যালাইড (বা সাবস্ট্রেট) উভয়ের ঘনমাত্রার ওপর নির্ভর করে, তাকে S_N2 বিক্রিয়া বলে।

বৈশিষ্ট্য : (i) 1° হ্যালাইডে S_N2 বেশি ঘটে, কারণ এতে স্টেরিক বাধা কম; (ii) বিক্রিয়াটি প্র-নিউক্লিওফাইল OH^- আয়নের ঘনমাত্রা বেশি থাকতে হয়; (iv) বিক্রিয়াটি ২য় ক্রমের হয় অর্থাৎ বিক্রিয়ানুযায়ী উভয়ের ঘনমাত্রার ওপর নির্ভর করে। (v) S_N2 বিক্রিয়াটিতে সরাসরি নিউক্লিওফাইল OH^- আয়ন বন্ধন গঠন ও পুরাতন বন্ধন ভাঙ্গনজনিত অবস্থান্তর ‘অস্থায়ী জটিল’ সৃষ্টি হয়। (vi) কার্বোনিয়াম আয়ন সৃষ্টি হয় না; (vii) RX অপূর্ণ ত্রিমাত্রিক গঠনে কার্বনের যে পার্শ্বে X পরমাণু থাকে 180° অবস্থানে নিউক্লিওফাইল আক্রমণ করে। তাই উৎপাদ অপূর্ণ জ্যামিতিক কাঠামো উল্টে হয়।

হয়।

S_N2 বিক্রিয়ার কৌশল : $R-\overset{\delta+}{CH_2}-\overset{\delta-}{X}$ এর X পরমাণু যে দিকে থাকে, তার বিপরীত দিক কার্বন পরমাণুকে নিউক্লিওফাইল ($:OH^-$) আক্রমণ করে। তখন নতুন সমযোজী বন্ধন গড়ন এবং পুরনো $C-X$ বন্ধনটি ভেঙ্গে গিয়ে হ্যালাইড (X^-) আয়ন মুক্ত হয়। এই সময়কারে ‘অস্থায়ী জটিল’ সৃষ্টি হয়। শেষে $C-X$ বন্ধনটি ভেঙ্গে গিয়ে হ্যালাইড (X^-) আয়ন মুক্ত হয়। C -পরমাণুর সাথে পূর্ণ সমযোজী বন্ধনে আবদ্ধ হয়ে পড়ে। তখন উৎপাদ অপূর্ণ কর্তৃক পরিবর্তিত হয়।
 তিন পরমাণু বা মূলক ঝড়ে উল্টানো ছাতার মতো উল্টে গিয়ে নতুন ত্রিমাত্রিক গঠন লাভ করে।
 কাঠামো পরিবর্তনকে বিজ্ঞানী পল ওয়ালাডেনের নামানুসারে ওয়ালাডেন ইন্ভার্সন (Walden inv)



সক্রিয়তার ক্রম হলো $1^\circ \text{RX} > 2^\circ \text{RX} > 3^\circ \text{RX}$ ।
এবং 2°RX এর বেলায় $\text{S}_{\text{N}}1$ ও $\text{S}_{\text{N}}2$ সমভাবে ঘটে।

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Hsc Organic Chemistry Suggestion

Ensheng Dong



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