

H.S.C Test Examination - 2013

Chemistry- 2nd Paper (Creative)

Sub. Code :

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Full Marks : 40

Time : 2 Hours 10 Minutes

[N.B- The figure in the right margin indicates full marks. Read the stems carefully and answer the associated questions. Answer any four questions.]

1. Two factories X and Y produce H_2SO_4 by contact process and chamber process. Experiment shows that Cu dissolves in the H_2SO_4 of factory X, but the H_2SO_4 of factory Y unable to dissolve Cu.

- What is 'Aqua Regia'? 1
- How will you identify Cu^{2+} ion in the solution? 2
- What happens when the H_2SO_4 of factory 'X' in the stem is heated mixing with NaI. 3
- What do you think, the step should be, to make the acid of factory 'Y' fit to dissolve Cu- explain. 4

2. The formula of an aromatic compound 'A' containing a substitute Y is

 $\ddot{Y}$. Due to the presence of group Y 'A' 'A' shows more reactivity than benzene in the electrophilic substitution reaction.

- What is free radical? 1
- Write down the name and formula of a homologue of CH_3-O-CH_3 . 2
- Describe the Fridel-Craft's reaction of compound 'A' with appropriate condition. 3
- How the radical $\ddot{Y}$ of the stem increase the reactivity of benzene ring- Explain. 4

3. The molecular formula of two Ketons 'A' and 'B' is $C_5H_{10}O$. Among than 'A' gives iodoform 'B' does give iodoform test.

- What is S_N^1 reaction? 1
- Why Zn is not a transition metal? 2
- Find out the name and structural formula of 'A' and 'B' with related reaction? 3
- Describe the reaction of B with NaOH solution mentioning the name of the reaction. 4

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