

সিলেট বোর্ড

বহুনির্বাচনি অধীক্ষা

তদারকন

প্রথম পত্র

[২০২৩ সালের সিলেটবাস অধ্যায়ী]

সময়—২৫ মিনিট

পূর্ণমান—২৫

বিষয় কোড :

1 7 6

সেট :

খ

[বিশেষ দ্রষ্টব্য : সরবরাহকৃত বহুনির্বাচনি অধীক্ষার উত্তরপত্রে প্রশ্নের ত্রমিক নম্বরের বিপরীতে প্রদত্ত বর্ণসম্বলিত বৃত্তসমূহ হতে সঠিক/সর্বোৎকৃষ্ট উত্তরের বৃত্তটি বলা পরেই কলাম দ্বারা সম্পূর্ণ করতে কর। প্রতিটি প্রশ্নের মান ১।]

প্রশ্নপত্রে কোনো প্রকার দাণ্ড/চিহ্ন দেয়া যাবে না।

- ১। চিকিৎসা বিজ্ঞানে ডিজিগবেষণিতে কোনটি ব্যবহার করা হয়?

(ক) X-ray
(খ) IR
(গ) MRI
(ঘ) UV

উল্লিখিত থেকে ২ এবং ৩ নং প্রশ্নের উত্তর দাও :

মৌল	বহির্গতরের ইলেকট্রন বিন্যাস	সকল থেকে n এর মান 2
A	$(n-1)s^1$	
B	ns^2np^4	
C	$(n+1)s^2(n+1)p^4$	

A, B এবং C প্রদত্ত মৌলের প্রতীক নয়।

- ২। A_2B ও A_2C যৌত অবস্থার ত্রিমাত্রিক ক্রম—

(ক) আয়নিক বন্ধন
(খ) সমযোজী বন্ধন
(গ) হাইড্রোজেন বন্ধন
(ঘ) দল্লিবেল বন্ধন

- ৩। A_2B যৌলের ক্ষেত্রে—

I. sp^3 সংকরণ ঘটে
II. বন্ধন কোণ 104.5°
III. $(L_p - L_p)$ বিকর্ষণ বিদ্যমান

নিচের কোনটি সঠিক?

(ক) I ও II
(খ) II ও III
(গ) I ও III
(ঘ) I, II ও III

- ৪। নিচের মৌলগুলোর ১ম আয়নীকরণ বিভবের সঠিক ক্রম কোনটি?

(ক) $Be > B > N > O$
(খ) $N > O > Be > B$
(গ) $O > N > B > Be$
(ঘ) $B > Be > N > O$

- ৫। হেলিনের প্রকৃত ভর কত?

(ক) $1.60 \times 10^{-24} g$
(খ) $1.66 \times 10^{-24} g$
(গ) $1.673 \times 10^{-24} g$
(ঘ) $1.675 \times 10^{-24} g$

- ৬। কোনটি প্রথম অক্সাইড?

(ক) NO
(খ) N_2O_5
(গ) N_2O_3
(ঘ) NO_2

- ৭। S-গ্রক মৌলের সংখ্যা কতটি?

(ক) 7
(খ) 12
(গ) 14
(ঘ) 16

- ৮। পরমাণুর ৩য় শক্তিস্তরে ঘূর্ণিত ইলেকট্রনের কৌণিক ভরবেগ এর মান কোনটি?

(ক) $\frac{nh}{2\pi}$
(খ) $\frac{3h}{2\pi}$
(গ) $\frac{2h}{4\pi}$
(ঘ) $\frac{9h}{2\pi}$

- ৯। কোনটি উভয়দিকী অক্সাইড?

(ক) Al_2O_3
(খ) Na_2O_2
(গ) KO_2
(ঘ) Fe_2O_3

উল্লিখিত হতে ১০ এবং ১১ নং প্রশ্নের উত্তর দাও :



- ১০। 200g NH_4Cl এর সম্পূর্ণ দ্রবককে $50^\circ C$ থেকে $25^\circ C$ তাপমাত্রায় ঠাণ্ডা করলে কত গ্রাম NH_4Cl ক্রিস্টালিত হবে?

(ক) 25
(খ) 50
(গ) 75
(ঘ) 90

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Arshad Iqbal



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properties of positive rays quantum numbers quantum theory relative abundance Rutherford model of atom shapes of orbitals solid iodine structure solids properties solvent extraction spectrometer spin quantum number states of matter stoichiometry sublimation thermometry scales types of solids unit cell van der Waals equation vapor pressure and spectrum

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thermal decomposition of group II carbonates thermal decomposition of group II nitrates uses of group II elements uses of group II metals uses of halogens and their compounds The Halogenoalkanes MCQ PDF e Book Chapter 14 practice test to solve MCQ questions on Halogenoalkanes uses of halogenoalkanes elimination reactions nucleophilic substitution in halogenoalkanes and nucleophilic substitution reactions The Hydrocarbons MCQ PDF e Book Chapter 15 practice test to solve MCQ questions on Introduction to alkanes sources of alkanes addition reactions of alkenes alkane reaction alkenes and formulas The Introduction to Organic Chemistry MCQ PDF e Book Chapter 16 practice test to solve MCQ questions on Organic chemistry functional groups organic reactions naming organic compounds stereoisomerism structural isomerism and types of organic reactions The Ionic Equilibria MCQ PDF e Book Chapter 17 practice test to solve MCQ questions on Introduction to ionic equilibria buffer solutions equilibrium and solubility indicators and acid base titrations pH calculations and weak acids The Lattice Energy MCQ PDF e Book Chapter 18 practice test to solve MCQ questions on Introduction to lattice energy ion polarization lattice energy value atomization and electron affinity Born Haber cycle and enthalpy changes in solution The Moles and Equations MCQ PDF e Book Chapter 19 practice test to solve MCQ questions on Amount of substance atoms molecules mass chemical formula and equations gas volumes mole calculations relative atomic mass solutions and concentrations The Nitrogen and Sulfur MCQ PDF e Book Chapter 20 practice test to solve MCQ questions on Nitrogen gas nitrogen and its compounds nitrogen and gas properties ammonia ammonium compounds environmental problems caused by nitrogen compounds and nitrate fertilizers sulfur and oxides sulfuric acid and properties and uses of sulfuric acid The Organic and Nitrogen Compounds MCQ PDF e Book Chapter 21 practice test to solve MCQ questions on Amides in chemistry amines amino acids peptides and proteins The Periodicity MCQ PDF e Book Chapter 22 practice test to solve MCQ questions on Acidic oxides basic oxides aluminum oxide balancing equation period 3 chlorides balancing equations reactions with chlorine balancing equations reactions with oxygen bonding nature of period 3 oxides chemical properties of chlorine chemical properties of oxygen chemical properties periodicity chemistry periodic table chemistry oxides chlorides of period 3 elements electrical conductivity in period 3 oxides electronegativity of period 3 oxides ionic bonds molecular structures of period 3 oxides oxidation number of oxides oxidation numbers oxides and hydroxides of period 3 elements oxides of period 3 elements period III chlorides periodic table electronegativity physical properties periodicity reaction of sodium and magnesium with water and relative melting point of period 3 oxides The Polymerization MCQ PDF e Book Chapter 23 practice test to solve MCQ questions on Types of polymerization polyamides polyesters and polymer deductions The Rates of Reaction MCQ PDF e Book Chapter 24 practice test to solve MCQ questions on Catalysis collision theory effect of concentration reaction kinetics and temperature effect on reaction rate The Reaction Kinetics MCQ PDF e Book Chapter 25 practice test to solve MCQ questions on Reaction kinetics catalysts kinetics and reaction mechanism order of reaction rate constant k and rate of reaction The Redox Reactions and Electrolysis MCQ PDF e Book Chapter 26 practice

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Table of Contents Hsc Chemistry 1st Paper Mcq

1. Understanding the eBook Hsc Chemistry 1st Paper Mcq
 - The Rise of Digital Reading Hsc Chemistry 1st Paper Mcq
 - Advantages of eBooks Over Traditional Books
2. Identifying Hsc Chemistry 1st Paper Mcq
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hsc Chemistry 1st Paper Mcq
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Chemistry 1st Paper Mcq

- Personalized Recommendations
 - Hsc Chemistry 1st Paper Mcq User Reviews and Ratings
 - Hsc Chemistry 1st Paper Mcq and Bestseller Lists
5. Accessing Hsc Chemistry 1st Paper Mcq Free and Paid eBooks
 - Hsc Chemistry 1st Paper Mcq Public Domain eBooks
 - Hsc Chemistry 1st Paper Mcq eBook Subscription Services
 - Hsc Chemistry 1st Paper Mcq Budget-Friendly Options
 6. Navigating Hsc Chemistry 1st Paper Mcq eBook Formats
 - ePub, PDF, MOBI, and More
 - Hsc Chemistry 1st Paper Mcq Compatibility with Devices
 - Hsc Chemistry 1st Paper Mcq Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hsc Chemistry 1st Paper Mcq
 - Highlighting and Note-Taking Hsc Chemistry 1st Paper Mcq
 - Interactive Elements Hsc Chemistry 1st Paper Mcq
 8. Staying Engaged with Hsc Chemistry 1st Paper Mcq
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hsc Chemistry 1st Paper Mcq
 9. Balancing eBooks and Physical Books Hsc Chemistry 1st Paper Mcq
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hsc Chemistry 1st Paper Mcq
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Hsc Chemistry 1st Paper Mcq
 - Setting Reading Goals Hsc Chemistry 1st Paper Mcq
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Hsc Chemistry 1st Paper Mcq

- Fact-Checking eBook Content of Hsc Chemistry 1st Paper Mcq
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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