

- Silver reacts with H_2S present in the atmosphere to form a black color compound Silver Sulphide. This phenomenon is called as corrosion.
- Black color compound formed is Silver Sulphide



39. On heating blue coloured powder of copper (II) nitrate in a boiling tube, copper oxide (black), oxygen gas and a brown gas X is formed

- Write a balanced chemical equation of the reaction.
- Identify the brown gas X evolved.
- Identify the type of reaction.
- What could be the pH range of aqueous solution of the gas X?

100



- (b) The brown gas X is nitrogen dioxide, NO_2 .
(c) Reaction involved is Thermal decomposition.
(d) $\text{pH} < 7$ because NO_2 dissolves in water to form acidic solution (pH range below 7).

40. Give the characteristic tests for the following gases

- (a) CO_2
(b) SO_2
(c) O_2
(d) H_2

- Pass CO_2 into limewater which will turn water into milky. This is the confirmation test for the presence of Carbon-di-oxide.
- Smell is the characteristic feature of SO_2 which smells like rotten egg.
- Test for oxygen involves burning of match stick near oxygen makes it burn even more brightly.
- When Burning matchstick is brought near H_2 gas, the flame burns with the pop sound. This is the test to confirm Hydrogen gas.

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Yijin Wang



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