

LONDON COLLEGE OF INTERNATIONAL CHEMISTS – 2020/2021

NAME: _____ COURSE: _____

ENTRY NUMBER: _____ ALPAC: _____

ICM: DECO (CONCEPTS IN ORGANIC CHEMISTRY)

ICM 11

Class : ICM-001-001

Group: (ICM- _____)

Date : _____

Time : 00:00 - 01:00

PART – I

20 x 2 = 40 marks

Answer ALL the questions.

1. State three bonding/intermolecular principles.
2. Describe hydrogen bonding in alcohols in the context of their boiling points.
3. Write the chemical formulae for CH_3OH in the H_2O .
4. Describe the structure of hydrogen bonding in H_2O molecules.
5. What intermolecular forces between molecules govern the boiling point of an alcohol and the rate of evaporation of an alcohol in comparison to the boiling point of an ether compound having same H_2O ?
6. Describe and explain how hydrogen bonding between water molecules affects the boiling point.
7. What are alcohols – definition and classification?
8. What are the uses of alcohols?
9. Describe the uses of alcohols: CH_3OH , $\text{C}_2\text{H}_5\text{OH}$, $\text{C}_3\text{H}_7\text{OH}$.
10. What are the uses of alcohols? Give 10 examples.

PART – II

20 x 2 = 40 marks

Answer any EIGHT questions.

11. What are alcohols and aldehydes? Write the chemical formulae and give their uses.
12. Write the uses of alcohols in the laboratory and in industry.
13. Describe the uses of alcohols in the laboratory and in industry.
14. Write the chemical formulae of alcohols and aldehydes.
15. What are alcohols and aldehydes? Give their uses.
16. Write the chemical formulae of alcohols and aldehydes.
17. Describe the uses of alcohols in the laboratory and in industry.
18. Describe the uses of alcohols in the laboratory and in industry.
19. Write the chemical formulae of alcohols and aldehydes.
20. Write the chemical formulae of alcohols and aldehydes.
21. Write the chemical formulae of alcohols and aldehydes.
22. Describe the uses of alcohols in the laboratory and in industry.
23. Describe the uses of alcohols in the laboratory and in industry.
24. Describe the uses of alcohols in the laboratory and in industry.
25. Describe the uses of alcohols in the laboratory and in industry.
26. Describe the uses of alcohols in the laboratory and in industry.
27. Describe the uses of alcohols in the laboratory and in industry.
28. Describe the uses of alcohols in the laboratory and in industry.
29. Describe the uses of alcohols in the laboratory and in industry.
30. Describe the uses of alcohols in the laboratory and in industry.

PART – III

20 x 2 = 40 marks

Answer any EIGHT questions.

31. What are alcohols and aldehydes? Write the chemical formulae and give their uses.
32. Write the uses of alcohols in the laboratory and in industry.
33. Describe the uses of alcohols in the laboratory and in industry.
34. Write the chemical formulae of alcohols and aldehydes.
35. What are alcohols and aldehydes? Give their uses.
36. Write the chemical formulae of alcohols and aldehydes.
37. Describe the uses of alcohols in the laboratory and in industry.
38. Describe the uses of alcohols in the laboratory and in industry.
39. Write the chemical formulae of alcohols and aldehydes.
40. Write the chemical formulae of alcohols and aldehydes.
41. Write the chemical formulae of alcohols and aldehydes.
42. Describe the uses of alcohols in the laboratory and in industry.
43. Describe the uses of alcohols in the laboratory and in industry.
44. Describe the uses of alcohols in the laboratory and in industry.
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47. Describe the uses of alcohols in the laboratory and in industry.
48. Describe the uses of alcohols in the laboratory and in industry.
49. Describe the uses of alcohols in the laboratory and in industry.
50. Describe the uses of alcohols in the laboratory and in industry.

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Michael Brown



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