

SECTION - I

Q. 1. Select and write the most appropriate answers from the given alternatives for each sub-question :

[7]

- (i) The Arrhenius equation is _____.
- (a) $k = A e^{\frac{RT}{E_a}}$ (b) $A = k e^{\frac{-E_a}{RT}}$
- (c) $k = A e^{\frac{-RT}{E_a}}$ (d) $k = A e^{\frac{-K_a}{RT}}$
- (ii) If the enthalpy of vaporisation of water at 100°C is 186.5 J . mol⁻¹, the entropy of vaporization will be _____.
- (a) 4.0 J . K⁻¹. mol⁻¹ (b) 3.0 J . K⁻¹. mol⁻¹
- (c) 1.5 J . K⁻¹. mol⁻¹ (d) 0.5 J . K⁻¹. mol⁻¹
- (iii) The atomicity of sulphur in orthorhombic sulphur is _____.
- (a) 8 (b) 6 (c) 4 (d) 2
- (iv) The major binding force in diamond is _____.
- (a) Covalent bond (b) Ionic bond
- (c) Metallic bond (d) Co-ordinate covalent bond
- (v) The boiling point of water at high altitude is low, because _____.
- (a) the temperature is low. (b) the atmospheric pressure is low.
- (c) the temperature is high. (d) the atmospheric pressure is high.
- (vi) The molar conductivity of cation and anion of salt BA are 180 and 220 mhos respectively. The molar conductivity of salt BA at infinite dilution is _____.
- (a) 90 mhos.cm². (b) 110 mhos.cm².mol⁻¹
- (c) 200 mhos.cm².mol⁻¹ (d) 400 mhos.cm².mol⁻¹
- (vii) What is the process in which concentrated ore is reduced to the corresponding metal by heating at high temperature with a reducing agent?
- (a) Polling (b) Pyrometallurgy (c) Hydrometallurgy (d) Calcination

Q. 2. Answer any THREE of the following :

[9]

- (i) Describe anomalous behaviour of oxygen as compared with other elements of group 16 with reference to –
- (a) Magnetic property (b) Oxidation state (c) Hydrides
- (ii) What is the value of $\Delta S_{\text{surroundings}}$ for the following reaction at 298 K –
- $$6\text{CO}_{2(g)} + 6\text{H}_2\text{O}_{(l)} \longrightarrow \text{C}_6\text{H}_{12}\text{O}_{6(s)} + 6\text{O}_{2(g)}$$
- Given that : $\Delta G^\circ = 2879 \text{ kJ mol}^{-1}$, $\Delta S = -210 \text{ JK}^{-1} \text{ mol}^{-1}$.
- (iii) Sucrose decomposes in acid solution to give glucose and fructose according to the first order rate law. The half life of the reaction is 3 hours. Calculate fraction of sucrose which will remain after 8 hours.
- (iv) A solution containing 0.73 g of camphor (molar mass 152 g . mol⁻¹) in 36.8 g of acetone (boiling point 56.3°C) boils at 56.55°C. A solution of 0.564 g of unknown compound in the same weight of acetone boils at 56.46°C. Calculate the molar mass of the unknown compound.

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**Yulan Lin, Waqas Sami, Mila Nu Nu
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Engaging in Educational Research Raqib Chowdhury, Mahbub Sarkar, Foez Mojumder, M Moninoor Roshid, 2018-10-01
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National Budget Framework Paper for the Financial Years ... Uganda, 2015

Active and Healthy Aging and Quality of Life: Interventions and Outlook for the Future Yulan Lin, Waqas Sami, Mila Nu Nu Htay, Sudip Bhattacharya, 2024-04-15
The continuous growth of older adult populations as a consequence of demographic changes is a huge challenge. To prevent costly and negative impacts effects on the population as a whole it is crucial that societies increase their knowledge of promoting good health among older adults to promote good health and a better quality of life in their later years. Active aging is the process of optimizing opportunities for health participation and security in order to enhance quality of life as people age. An active and healthy life has remained one of the most important aspirations for all people both young and older adults alike. This ambition has become a genuine possibility for many due to a rising life expectancy among people of diverse attributes across the world. While celebrating longer life and with more financial security in later life than ever before we need to challenge how these aspirations can be sustained through our own behavioural responses and through public policy institutional reforms and innovations. The challenge is to identify recommend and promote strategies and interventions that stimulate and sustain the activity independence and health of people of all ages especially older adults and in the process promote the well being and quality of life of people and make public welfare systems more sustainable.

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 - The Rise of Digital Reading Hsc Ques Papers 2014 Oct Target
 - Advantages of eBooks Over Traditional Books
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 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hsc Ques Papers 2014 Oct Target
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Ques Papers 2014 Oct Target
 - Personalized Recommendations
 - Hsc Ques Papers 2014 Oct Target User Reviews and Ratings
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- 8. Staying Engaged with Hsc Ques Papers 2014 Oct Target
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hsc Ques Papers 2014 Oct Target
- 9. Balancing eBooks and Physical Books Hsc Ques Papers 2014 Oct Target
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hsc Ques Papers 2014 Oct Target
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Hsc Ques Papers 2014 Oct Target
 - Setting Reading Goals Hsc Ques Papers 2014 Oct Target
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hsc Ques Papers 2014 Oct Target
 - Fact-Checking eBook Content of Hsc Ques Papers 2014 Oct Target
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- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

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- Interactive and Gamified eBooks

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