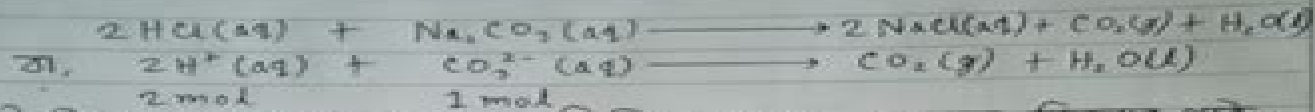


NAME OF THE EXPERIMENT: 0.1M Na_2CO_3	Page No.	
উদ্যম দ্বারা সংকলিত নমুনা উদ্যম (HCl/উদ্যম)	Expt No.	01
চলমান দিন	Date :	

মূলনীতি:

Na_2CO_3 ও HCl এর প্রকৃষ্ট বিক্রিয়া সমীকরণ নিম্নরূপ:



টাইট্রেশনের তৈরিতে, প্রকৃষ্ট বিক্রিয়া কোষ প্রকৃষ্ট বিক্রিয়ায় পরিণত হয়, CO_3^{2-} এর মোল সংখ্যা (ফ্রাক্ট) $\equiv \text{H}^+$ এর মোল সংখ্যা (ক্যালি ব্রোমিডে দেয়া)

প্রয়োজনীয় সামগ্রিক তথ্য:

- 1) 0.1M Na_2CO_3 উদ্যম
- 2) 0.1M HCl উদ্যম
- 3) নির্দেশক : মিথাইল অরেঞ্জ

প্রয়োজনীয় যন্ত্রপাতি:

- 1) কনিক্যাল ফ্লাস্ক
- 2) ব্যালো
- 3) পিপেট
- 4) ক্যাম্পাস মেসার

কার্যের ধারা:

- 1) প্রস্তুত করা 0.1M HCl উদ্যম (প্রকৃষ্ট উদ্যম) অংশ নমুনা HCl দ্বারা ব্যালোটি বর্ণ করে এবং মেসারের মাধ্যমে ক্যাম্পাস ফ্লাস্ক বাফারের আউটপুট নিই।
- 2) প্রস্তুত করা 0.1M Na_2CO_3 উদ্যম (প্রকৃষ্ট উদ্যম) এর আনুমানিক ফ্রাক্ট থেকে পিপেট দ্বারা 10ml উদ্যম কনিক্যাল ফ্লাস্কে নিই, কনিক্যাল ফ্লাস্কে উদ্যম 1-2 ফোটা মিথাইল অরেঞ্জ যোগ করে, উদ্যম উদ্যমের বর্ণ হলুদ হয়।
- 3) কনিক্যাল ফ্লাস্কে তৈরি একটি সাদা সাদা টুকরা রাখি, ব্যালোটি HCl উদ্যম উদ্যমের মাধ্যমে নিয়ে পিপেট দ্বারা '১ম পাইল' - এ ফেরে করে, চিহ্নিত করা সাদা ব্যালোটির বর্ণ লক্ষ্য করে ব্যালোটির HCl উদ্যম ফোটা ফোটা কনিক্যাল ফ্লাস্কে যোগ করে

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Yicheng Fang



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