

Chapter 1 Review

1. Name a line that contains points  $T$  and  $P$ .

line  $g$

2. Name the plane that contains  $\overline{TN}$  and  $\overline{QR}$ .

plane  $S$

3. Name three collinear points.

$S, X, M$

4. Name four noncoplanar points.

$S, X, R, W$

5. Find the length of the object.



4.2



7. Find the value of  $x$  and  $KL$  if  $K$  is between  $J$  and  $L$ .

$JK = 2x$ ,  $KL = x + 2$ , and  $JL = 5x - 10$

$$\begin{array}{c} J \quad 2x \quad K \quad x+2 \quad L \\ \hline 5x-10 \end{array}$$

$$\begin{aligned} 3x + 2 &= 5x - 10 \\ 12 &= 2x \\ \boxed{6 = x} \end{aligned}$$

8. Jorge used the figure at the right to make a pattern for a mosaic he plans to inlay on a tabletop. Name all of the congruent segments in the figure.

$$\overline{FA} \cong \overline{AB} \cong \overline{ED} \cong \overline{DC} \quad , \quad \overline{FE} \cong \overline{BC}$$



9. Find the distance between each pair of points.



$$\sqrt{7^2 + 4^2} = \sqrt{65}$$

b.  $C(-2, -1)$ ,  $K(8, 3)$

$$\begin{aligned} x^2 &= 10 \\ y^2 &= 4 \end{aligned}$$

$$\sqrt{4^2 + 10^2} = \sqrt{116}$$

10. Find the coordinates of the midpoint of a segment with the given endpoints.

a.  $K(-9, 3)$ ,  $H(5, 7)$

$$= \frac{-9}{2}, \frac{10}{2}$$

$$(-2, 5)$$

b.  $W(-12, -7)$ ,  $T(-8, -4)$

$$= \frac{-20}{2}, \frac{-11}{2}$$

$$(-10, -5\frac{1}{2})$$

# Geometry Practice Masters Answer Guide

**Karin Nielsen-Saines**



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oxide and nitric acid magnesium oxide nitric acid magnesium nitrate

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type of chemical reaction for this reaction we have a neutralization reaction balancing strategies here we have a  
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solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the  
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web nov 20 2020 in this video we ll balance the equation  $\text{mg} + \text{hno}_3 \rightarrow \text{mg}(\text{no}_3)_2 + \text{no}_2 + \text{h}_2\text{o}$  and provide the correct coefficients for each compound to balance  $\text{mg} + \text{hno}_3 \rightarrow \text{mg}(\text{no}_3)_2 + \text{no}_2 + \text{h}_2\text{o}$  you ll need to be

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web magnesium hydroxide nitric acid balanced total equation  $\text{mg}(\text{oh})_2(\text{aq}) + 2\text{hno}_3(\text{aq}) \rightarrow \text{mg}(\text{no}_3)_2(\text{aq}) + 2\text{h}_2\text{o}$  total ionic equation  $\text{mg}^{2+}(\text{aq}) + 2\text{oh}^{-}(\text{aq}) + 2\text{h}^{+}(\text{aq}) + 2\text{no}_3^{-}(\text{aq}) \rightarrow \text{mg}^{2+}(\text{aq}) + 2\text{no}_3^{-}(\text{aq}) + 2\text{h}_2\text{o}$  net ionic equation  $2\text{oh}^{-}(\text{aq}) + 2\text{h}^{+}(\text{aq}) \rightarrow 2\text{h}_2\text{o}$  sulfuric acid and lithium hydroxide balanced total equation  $\text{h}_2\text{so}_4(\text{aq}) + 2\text{lioh}(\text{aq}) \rightarrow \text{h}_2\text{o}$