

Book 3: Example questions. All questions were translated from German. Keys are marked with an asterisk [*]. See text for details.

Question 1. Example of a kypnose-type question.

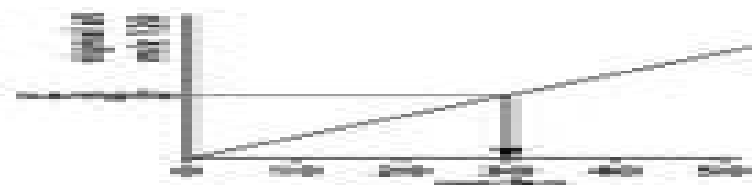
What increases the spectral resolution of an ICP-MS instrument?

- ☐ True ☐ False
- ☐ a Scavenger slit in a sector field instrument.
- ☐ a Using a dual mode detector with analog and pulse counting.
- ☐ a Detection of barium at m/z 138 instead of 137.
- ☐ a Increase the length of the quadrupole for an ICP-QMS.

Question 2. Silver has a mass of 107.87 g/mol. At which m/z do you detect silver with ICP-MS?

- ☐ A 106
- ☐ B* 107
- ☐ C 108
- ☐ D* 109

Question 3. You quantify tin in a sample with ICP-MS and external calibration at m/z 120. The isotope abundance of ^{120}Sn is about 33%. The calibration for the signal at m/z 120 indicates a mass fraction of 30 $\mu\text{g}/\text{kg}$ [see figure below]. What is the mass fraction of tin in the measurement aliquot?



- ☐ A 90 $\mu\text{g}/\text{kg}$
- ☐ B 60 $\mu\text{g}/\text{kg}$
- ☐ C* 30 $\mu\text{g}/\text{kg}$
- ☐ D 10 $\mu\text{g}/\text{kg}$

Question 4. Chocolate was digested in a closed vial under pressure using nitric acid and hydrogen peroxide using a microwave system. What constitutes the main matrix of the analyte cadmium in the prepared solution after the digestion of chocolate?

- ☐ A cadno., sugar, fat
- ☐ B* water
- ☐ C cadmium ions
- ☐ D no matrix

Question 5. How do you prepare a sample of salmon to differentiate between and quantify organic mercury species?

- ☐ A microwave digestion using HCl and H_2O_2
- ☐ B microwave digestion using HNO_3 and H_2O_2
- ☐ C microwave digestion using HNO_3
- ☐ D* no digestion

Multiple Choice Question For Instrumental Chemistry

Mark Steyvers



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