

The fundamental ideas in chemistry

1.	electron nucleus neutron	each for 1 mark		
				[3]
2.	(a) (i) B		1	
	(ii) D		1	
	(b) A and B – only one type of atom		1	
	C and D – more than one type of atom		1	
	accept element for atom			
	ignore the word 'mixture'			
	(chemically) bonded		1	
	accept (chemically) joined or similar idea of joined			
				[5]
3.	B carbon monoxide		1	
	CO	accept carbon oxide do not credit carbon dioxide do not credit if any superscripts or subscripts used but accept C ₂ O ₂ accept OC do not credit if obviously lower case	1	
	C water		1	
	H ₂ O	accept hydrogen oxide do not accept hydrogen hydroxide do not credit if obviously lower case or if 2 not subscript do not accept HOH accept OH ₂	1	
	D ammonia	do not accept ammonium	1	
	NH ₃	do not credit if obviously lower case, or if 3 not subscript accept nitrogen hydride or hydrogen nitride do not accept hydrogen nitrate or nitrite allow H ₃ N	1	
				[6]
4.	• correct reactants (i.e. sodium + water either way round)			
	• correct products (i.e. sodium hydroxide + hydrogen, either way round)			
	• arrow → / =	do not allow produce/makes or similar do not allow symbols or formulae		
		each for 1 mark		
				[3]
5.	correct use of 'react'/'reaction'/'reactants'/'combine' (not mixed/added/join)		3	
	correct use of 'produce'/'products'/'gives'/'forms'/'makes'/'creates' reactants			
	correctly identified			
	each for 1 mark			
	products correctly identified		1	

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Jicheng Xie



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Table of Contents May 13 C1 Markscheme

1. Understanding the eBook May 13 C1 Markscheme
 - The Rise of Digital Reading May 13 C1 Markscheme
 - Advantages of eBooks Over Traditional Books
2. Identifying May 13 C1 Markscheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an May 13 C1 Markscheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from May 13 C1 Markscheme
 - Personalized Recommendations
 - May 13 C1 Markscheme User Reviews and Ratings
 - May 13 C1 Markscheme and Bestseller Lists
5. Accessing May 13 C1 Markscheme Free and Paid eBooks
 - May 13 C1 Markscheme Public Domain eBooks

- May 13 C1 Markscheme eBook Subscription Services
- May 13 C1 Markscheme Budget-Friendly Options
- 6. Navigating May 13 C1 Markscheme eBook Formats
 - ePub, PDF, MOBI, and More
 - May 13 C1 Markscheme Compatibility with Devices
 - May 13 C1 Markscheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of May 13 C1 Markscheme
 - Highlighting and Note-Taking May 13 C1 Markscheme
 - Interactive Elements May 13 C1 Markscheme
- 8. Staying Engaged with May 13 C1 Markscheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers May 13 C1 Markscheme
- 9. Balancing eBooks and Physical Books May 13 C1 Markscheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection May 13 C1 Markscheme
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine May 13 C1 Markscheme
 - Setting Reading Goals May 13 C1 Markscheme
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of May 13 C1 Markscheme
 - Fact-Checking eBook Content of May 13 C1 Markscheme
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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