

CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

*Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B,

Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

Gas constant $R = 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1}$

$= 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}$

$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J}$ $1 \text{ Faraday} = 96500 \text{ coulombs}$.

Avogadro's number $= 6.023 \times 10^{23}$.

SECTION A – 14 MARKS

Question 1

- (A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1]
brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into _____.
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex $\text{K}_4[\text{Fe}(\text{CN})_6]$ are _____ and _____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives _____ and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

Turn over

Jnv 2014 Class 1chemistry Question Paper

Kathleen Armour



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Table of Contents Jnv 2014 Class 1chemistry Question Paper

1. Understanding the eBook Jnv 2014 Class 1chemistry Question Paper
 - The Rise of Digital Reading Jnv 2014 Class 1chemistry Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Jnv 2014 Class 1chemistry Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jnv 2014 Class 1chemistry Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jnv 2014 Class 1chemistry Question Paper
 - Personalized Recommendations
 - Jnv 2014 Class 1chemistry Question Paper User Reviews and Ratings
 - Jnv 2014 Class 1chemistry Question Paper and Bestseller Lists

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 - Jnv 2014 Class 1chemistry Question Paper Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Jnv 2014 Class 1chemistry Question Paper Compatibility with Devices
 - Jnv 2014 Class 1chemistry Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jnv 2014 Class 1chemistry Question Paper
 - Highlighting and Note-Taking Jnv 2014 Class 1chemistry Question Paper
 - Interactive Elements Jnv 2014 Class 1chemistry Question Paper
8. Staying Engaged with Jnv 2014 Class 1chemistry Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Jnv 2014 Class 1chemistry Question Paper
9. Balancing eBooks and Physical Books Jnv 2014 Class 1chemistry Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jnv 2014 Class 1chemistry Question Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Jnv 2014 Class 1chemistry Question Paper
 - Setting Reading Goals Jnv 2014 Class 1chemistry Question Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Jnv 2014 Class 1chemistry Question Paper
 - Fact-Checking eBook Content of Jnv 2014 Class 1chemistry Question Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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