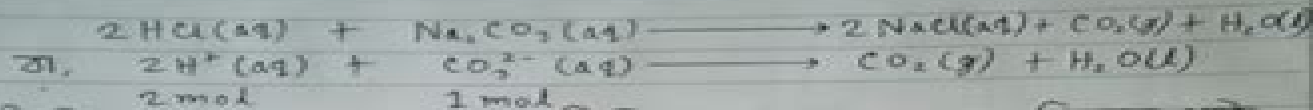


NAME OF THE EXPERIMENT: 0.1M Na_2CO_3	Page No.	
উদ্যম দ্বারা সংকলিত নমুনা উদ্যম (HCl/উদ্যম)	Expt No.	01
চলমান দিন	Date :	

মূলনীতি: Na_2CO_3 ও HCl এর প্রকৃত বিক্রিয়া সমীকরণ নিম্নরূপ:



টাইট্রেশনের তৈরিতে, প্রকৃত বিক্রিয়া কোথায় প্রকৃত বিক্রিয়া পাবে, CO_3^{2-} এর মোল সংখ্যা (ফ্রাক্ট) $\equiv \text{H}^+$ এর মোল সংখ্যা (ক্যালি ব্রোমিডে দেয়া)

প্রয়োজনীয় সামগ্রিক বস্তু:

- ১) 0.1M Na_2CO_3 দ্রব
- ২) 0.1M HCl দ্রব
- ৩) নির্দেশক : মিথাইল অরেঞ্জ

প্রয়োজনীয় যন্ত্রপাতি:

- ১) কনিক্যাল ফ্লাস্ক
- ২) ব্যালো
- ৩) পিপেট
- ৪) ক্যাম্পমই মেটার

কার্যের ধারা:

- ১) প্রস্তুত করা 0.1M HCl দ্রব (প্রকৃত দ্রব) অথবা নমুনা HCl দ্বারা ব্যালোটি বর্ণ করে এবং মেটারের মাধ্যমে ক্যাম্পমই মেটার দ্বারা পরিমাপ করে নেওয়া হবে।
- ২) প্রস্তুত করা 0.1M Na_2CO_3 দ্রব (প্রকৃত দ্রব) এর আনুমানিক ফ্রাক্ট দেওয়া পিপেট দ্বারা 10ml দ্রব কনিক্যাল ফ্লাস্কে নেওয়া হবে, কনিক্যাল ফ্লাস্কে দ্রব 1-2 ফোটা মিথাইল অরেঞ্জ যোগ করে, তখন দ্রবের বর্ণ হলুদ হবে।
- ৩) কনিক্যাল ফ্লাস্কে তৈরি একটি সাদা কাগজ টুকরা রাখা, ব্যালোটি HCl এর দ্রবের দিকে নিয়ে যোগ করে 'গম পাঠ' - এ ফেরত করে, চিহ্নিত করা সাদা ব্যালোটিতে বর্ণ লগ্ন হলে ব্যালোটিতে HCl দ্রব ফোটা ফোটা কনিক্যাল ফ্লাস্কে যোগ করে

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Daniela Niemeyer



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