

সুতরাং, বিশেষ MY_2 অম্লজিহ্ন হবে।



↑↑↑↑↑ ivlv fv+jv:

কিছু ভ্রাম্যমাণের কোনো প্রবণে একটি লবণ প্রস্তুত থাকবে যদি অম্লজিহ্ন হবে তা নির্ভর করে লবণটির প্রাকৃতিক তলপাতের উপর। যদি প্রবণে কোনো লবণের উপস্থিতি অম্লজিহ্নের আয়নিক তলপাতের মান লবণটির প্রাকৃতিক তলপাতের মানকে অতিক্রম করতে না পারে তবে লবণটি প্রবণে প্রস্তুত করা যাবে। আর যদি লবণের আয়নিক তলপাতের মান (K_{sp}) লবণটির প্রাকৃতিক তলপাতের মান (K_{sp}) কে অতিক্রম করে তবে লবণটি প্রবণ থেকে অম্লজিহ্ন হয়।

অর্থাৎ,

$K_{sp} < K_{sp}$: প্রবণটি অসম্পূর্ণ প্রবণ

$K_{sp} = K_{sp}$: প্রবণটি সম্পূর্ণ প্রবণ

$K_{sp} > K_{sp}$: প্রবণ থেকে প্রব অম্লজিহ্ন হয়

III A পাত্রে PN প্রবণ তৈরি করলে তা নিম্নলিখিত অম্লজিহ্ন হয়,



সুতরাং, PN প্রবণের ঘনমাত্রা, $0.1M$ হবে-

P^+ আয়নের ঘনমাত্রা, $[P^+] = 0.1 M$

N^- আয়নের ঘনমাত্রা, $[N^-] = 0.1 M$

এখানে, PN প্রবণে N^- আয়ন উপস্থিত। অন্যদিকে A পাত্রে MIN_3 লবণের একটি অম্লজিহ্ন N^- উপস্থিত থাকবে। A পাত্রে PN প্রবণ তৈরি করলে সমসাময়িক প্রতিক্রিয়া করলে MIN_3 এর প্রাকৃতিক তলপাত হবে।

যদি, MIN_3 এর লবণ প্রাকৃতিক তলপাত, S

প্রবণে MIN_3 নিম্নলিখিত অম্লজিহ্ন হয়,



সুতরাং, M^{3+} আয়নের ঘনমাত্রা, $[M^{3+}] = S$

N^- আয়নের ঘনমাত্রা, $[N^-] = 3S$

সেহেতু, N^- আয়ন PN ও MIN_3 উভয় দিকের উপস্থিত। সেহেতু,

N^- আয়নের মোট ঘনমাত্রা হবে, $[N^-] = (3S + 0.1)M$

এখন MIN_3 এর আয়নিক তলপাত,

$$K_{sp} = [M^{3+}] \times [N^-]^3$$

$$\begin{aligned} \text{কি, } 2.7 \times 10^{-11} &= S \times (3S + 0.1)^3 \quad [\because \text{কি, } K_{sp} = 2.7 \times 10^{-11}] \\ &= S \times [(3S)^3 + 3 \times (3S)^2 \times 0.1 + 3 \times 3S \times (0.1)^2 + (0.1)^3] \\ &= S \times [27S^3 + 2.7S^2 + 0.09S + 0.001] \\ &= 27S^4 + 2.7S^3 + 0.09S^2 + 0.001S \end{aligned}$$

$$\text{কি, } 2.7 \times 10^{-11} = 0.001S \quad [\because S^4, S^3, S^2 \text{ এর মান খুবই নগণ্য}]$$

$$\begin{aligned} \therefore S &= \frac{2.7 \times 10^{-11}}{0.001} \\ &= 2.7 \times 10^{-8} \text{ mol L}^{-1} \end{aligned}$$

অর্থাৎ, MIN_3 দিকের লবণ প্রাকৃতিক তলপাত, $2.7 \times 10^{-8} \text{ mol L}^{-1}$

অতএব, A পাত্রে PN প্রবণ তৈরি করলে MIN_3 এর প্রাকৃতিক তলপাত $10^{-8} \text{ mol L}^{-1}$ থেকে বেশি $2.7 \times 10^{-8} \text{ mol L}^{-1}$ হবে।

Hsc Chemistry Note Bd

M Woodhall



Hsc Chemistry Note Bd:

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