



EXAMPLE OF AN AUTOBIOGRAPHY

Growing up, I was always obsessed with beauty. I was born in a family with five other siblings, and I was one of two girls. My mother always taught us that as girls, we should always strive to be the best we could be, and that meant excelling in academics, sports, as well as beauty. My mother would sign up my sister and I for pageants constantly, but always ensuring that they were talent pageants and not simply beauty ones. "Beauty without brains is just one step away from the trash heap," my mother would always say. However, my clothes would never fit me quite right, and I remember always feeling very awkward next to my beautiful sister, who would always be the one to carry home the award.

One day, I decided that I did not want to wear the same outfit that my sister did. The day before the pageant, I snuck into my mother's closet and altered my outfit throughout the night, clumsily using glue and thread to try and make my outfit something that represented what I would want to wear. The next day during the competition, I saw my mother's look of horror as she realized that I had completely butchered the outfit, but the judges found it to be artistic and creative. From that point on, I started designing my own clothes, made for my body shape and made to my own tastes.

I started my own line of clothing when I was sixteen, and I have been in fashion design ever since. My goal is to design timeless pieces which give room for women to make alterations and customizations to suit their particular dimensions. No two people are the same, and I believe that no piece of clothing should fit two people the same way either. As a fashion designer, rebellion and unique expression are at the key of what I strive to create, and I will continue to do so with every new release of clothing that I design.

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Tobias Bleicker



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web apr 7 2014 molarity moles of solute / liters of solution 8.422 a first convert 250 ml to liters 250 / 1000 = 0.25 then calculate molarity 5 moles / 0.25 liters = 20 M 3. c a solution with molarity 2 requires 2 M of NaOH per liter so 4 x 2.8 M = 11.2 M a solution of molarity 1.5 M requires 1.5 mol of NaOH to every litre of solvent

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web answer the correct option is c explanation the molarity of a mixture can be calculated using the following formula $M_{\text{mix}} = \frac{M_1 V_1 + M_2 V_2}{V_1 + V_2}$ $\frac{1.5 \times 480 + 2 \times 520}{480 + 520} = 1.72 \text{ M}$ 5. with increase in temperature which of these changes a molality b fraction of solute present in water

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web 1. calculate the molarity of a solution which contains 0.40 mol of a substance dissolved in 1.6 L of a solution 2. what is the molarity of a solution containing 325 g of NaCl dissolved in 750 ml of solution 3. 140 g of KCl is dissolved in 600 ml of water what is the molarity 4. 724.4 g of ammonium phosphate in 4500 ml of alcohol

6.1 calculating molarity problems chemistry libretexts - Jun 13 2023

web determine the molarity for each of the following solutions 0.444 mol of COCl_2 in 0.654 L of solution 98.0 g of phosphoric

acid h 3 po 4 in 1 00 l of solution 0 2074 g of calcium hydroxide ca oh 2 in 40 00 ml of solution 10 5 kg

16 11 molality chemistry libretexts - Aug 15 2023

web molality differs from molarity only in the denominator while molarity is based on the liters of solution molality is based on the kilograms of solvent concentrations expressed in molality are used when studying properties of solutions related to vapor pressure and temperature changes

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