

Guided Reader- Honors Chemistry

Directions: Please answer the questions in the guided reading activity as you read chapter 9 of your textbook. A list of advice, with page numbers for your answers so you can reference them later.

1. Molecules have shapes and sizes that are defined by the _____ and distances between the _____ of their component atoms.
2. The shape of a molecule affects its _____, since the nose has receptor sites into which the molecule may fit and send impulses to the brain.
3. The overall shape of a molecule is determined by its _____.
4. Molecules with single central atoms conform to the general formula of AB_n . What does this mean?
5. The shape of AB_n molecules depends on the value of _____.
6. How many possible shapes are there for AB_2 molecules? Name them.
7. How many possible shapes are there for AB_3 molecules? Name them.
8. A _____ of electrons defines a region in which the electrons will most likely be found.
9. A _____, also known as _____ of electrons defines an electron domain that is located principally on one atom.
10. How many bonding pairs of electrons are around the central atom in a molecule of ammonia?
11. How many nonbonding pairs of electrons are around the central atom in a molecule of ammonia?
12. How many electron domains does are around the central atom in a molecule of ammonia?
13. A multiple bond in a molecule constitutes a _____ electron domain.
14. List the three things that can produce an electron domain around the central atom.
15. What is the best arrangement of a given number of electron domains?
16. Complete the following table:

# of electron domains	Electron domain geometry	Predicted bond angle
2		
3		
4		
5		
6		
17. What is the difference between electron-domain geometry and molecular geometry?
18. When all electron domains in a molecule arise from bonds, the molecular geometry is _____ to the electron-domain geometry.

Guided Reading Activities For Civics Today Chapter 9

D Kirk



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